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                 CHEMLIST enhanced with new search and display field
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                 JAPIO enhanced with IPC 8 features and functionality
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         NOV 10
                 CA/CAplus F-Term thesaurus enhanced
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         NOV 20
                 CAS Registry Number crossover limit increased to 300,000 in
                 additional databases
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      9
         NOV 20
                 CA/CAplus to MARPAT accession number crossover limit increased
                 to 50,000
NEWS 10
         DEC 01
                 CAS REGISTRY updated with new ambiguity codes
NEWS 11
         DEC 11
                 CAS REGISTRY chemical nomenclature enhanced
NEWS 12
         DEC 14
                 WPIDS/WPINDEX/WPIX manual codes updated
NEWS 13
         DEC 14
                 GBFULL and FRFULL enhanced with IPC 8 features and
                 functionality
NEWS 14
         DEC 18
                 CA/CAplus pre-1967 chemical substance index entries enhanced
                 with preparation role
NEWS 15
         DEC 18
                 CA/CAplus patent kind codes updated
NEWS 16
         DEC 18
                 MARPAT to CA/CAplus accession number crossover limit increased
                 to 50,000
NEWS 17
         DEC 18
                 MEDLINE updated in preparation for 2007 reload
NEWS 18
         DEC 27
                 CA/CAplus enhanced with more pre-1907 records
NEWS 19
         JAN 08
                 CHEMLIST enhanced with New Zealand Inventory of Chemicals
NEWS EXPRESS
              NOVEMBER 10 CURRENT WINDOWS VERSION IS V8.01c, CURRENT
              MACINTOSH VERSION IS V6.0c(ENG) AND V6.0Jc(JP),
              AND CURRENT DISCOVER FILE IS DATED 25 SEPTEMBER 2006.
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NEWS X25
              X.25 communication option no longer available
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<---->
SEARCH ENDED BY USER
=> s ((bioactive glass) or ceramic or bioglass or (bioactive ceramic))
        781795 ((BIOACTIVE GLASS) OR CERAMIC OR BIOGLASS OR (BIOACTIVE CERAMIC)
=> s l1 and ((calcium oxide) or CaO)
L2
         31755 L1 AND ((CALCIUM OXIDE) OR CAO)
=> s 12 and (silica or SiO2)
        17778 L2 AND (SILICA OR SIO2)
=> s 13 and ((boron oxide) or B2O3)
          3885 L3 AND ((BORON OXIDE) OR B2O3)
=> s L4 and ((magnesium oxide) or MgO)
          2589 L4 AND ((MAGNESIUM OXIDE) OR MGO)
=> s L5 and ((calcium floride) or CaF2))
UNMATCHED RIGHT PARENTHESIS 'CAF2))'
The number of right parentheses in a query must be equal to the
number of left parentheses.
=> s L5 and ((calcium floride) or CaF2)
          111 L5 AND ((CALCIUM FLORIDE) OR CAF2)
=> s L6 and ((phosphorus pentoxide) or P2O5)
L7
            58 L6 AND ((PHOSPHORUS PENTOXIDE) OR P2O5)
=> s 17 and biodegrad?
             3 L7 AND BIODEGRAD?
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=> d 18 1-3 ibib abs

L8 ANSWER 1 OF 3 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2004:182250 CAPLUS

DOCUMENT NUMBER: 140:205207

TITLE: Biodegradable and bioactive

glass-ceramics, and method for fabricating the

same

INVENTOR(S): Yu, Hyun Seung; Hong, Kug Sun; Kim, Hwan; Lee, Dong

Ho; Lee, Choon Ki; Chang, Bong Soon; Kim, Deug Joong;

Seo, Jun Hyuk; Lee, Jae Hyup; Park, Ki Soo

PATENT ASSIGNEE(S): S. Korea

SOURCE: U.S. Pat. Appl. Publ., 17 pp.

CODEN: USXXCO

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 2004043053	A1	20040304	US 2003-648217	20030827
KR 2004021111	Α	20040310	KR 2002-52566	20020902
PRIORITY APPLN. INFO.:			KR 2002-52566 A	20020902

AB Disclosed herein is a biodegradable and bioactive

glass-ceramic fabricated by mixing a slowly biodegradable glass-ceramic and a highly

biodegradable glass-ceramic in a predetd. mixing ratio

wherein the bioactivity is maintained to be constant, and the biodegrdn.

rate is controlled by the mixing ratio. The biodegradable and

bioactive glass-ceramic is fabricated from a composition consisting of calcium oxide (CaO),

silica (SiO2), boron oxide (B2O3), magnesium oxide (MgO),

calcium fluoride (CaF2) and phosphorus

pentoxide (P2O5). Preparation of a biodegradable

and bioactive glass-ceramic according to

above method is disclosed. Electron microscopic images shown hydroxycarbonated apatite layers were formed on the entire surfaces of the specimens taken out 1 day after soaking in simulated body fluid thus suggesting that the specimens were highly bioactive. As the content of B203 in the specimens increased, the weight of the specimens was greatly reduced. This demonstrated that the biodegrdn. of the specimens had actively proceeded.

L8 ANSWER 2 OF 3 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER: 2003:133768 EPFULL

ENTRY DATE PUBLICATION: 20051228
UPDATE DATE PUBLICAT:: 20061206
DATA UPDATE DATE: 20061206
DATA UPDATE WEEK: 200649

TITLE (ENGLISH): BIOCOMPATIBLE MATERIAL TITLE (FRENCH): MATERIAU BIOCOMPATIBLE TITLE (GERMAN): BIOKOMPATIBLES MATERIAL

INVENTOR(S): Martinez Fernandez, Julian, Universidad de

Sevilla, Pabellon de Brasil s/n, 41012 Sevilla, ES; Ramirez de Arellano Lopez, Antonio, Universidad de Sevilla, Pabellon de Brasil s/n, 41012 Sevilla, ES; Varela Feria, Francisco Manuel, Universidad de Sevilla, Pabellon de Brasil s/n, 41012 Sevilla, ES; Gonzalez Fernandez, Pio Manuel, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES;

Serra Rodriguez, Julia A., Uni. de Vigo,ETS de

Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Liste Carmueja, Sara, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Chiussi, Stefano, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Perez Borrajo, Jacinto, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Arias Otero, Jose Luis, Uni. de

Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Leon Fong, Betty, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Perez-Martinez Y Perez-Amor, Mariano, Uni. de Vigo, ETS

de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES

PATENT APPLICANT(S): Universidad de Sevilla,, OTRI - Universidad de Sevilla,

c/ Valparaiso, 5 1 planta, 41013 Sevilla, ES; Universidad de Vigo, Officina I+D - Universidad de Vigo, Edificio CACTI, Campus Universitario de Vigo,

36200 Vigo (Pontevedra), ES

PATENT APPL. NUMBER: 2431773; 1896640

AGENT: Temino Ceniceros, Ignacio, Abril Abogados, Amador de

los Rios, 1 1, 28010 Madrid, ES

AGENT NUMBER: 159233
DOCUMENT TYPE: Patent
LANGUAGE OF FILING: Spanish
LANGUAGE OF PUBL: English
LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE: German; English; French

PATENT INFO TYPE: EPA1 Application published with search report

PATENT INFORMATION: PATENT INFORMATION:

 NUMBER
 KIND
 DATE

 NUMBER
 KIND
 DATE

 EP 1609441
 A1 20051228

 WO 2004056292
 20040708

DESIGNATED STATES: AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HU IE IT LI

LU MC NL PT RO SE SI SK TR

APPLICATION INFO.: EP 2003-785961 A 20031216 WO 2003-ES638 A 20031216 PRIORITY INFO.: ES 2002-3052 A 20021220

### ABEN

The invention relates to a novel biocompatible material which is intended to be used in the production of implants, prostheses or biomedical devices, comprising biomorphic SiC ceramics as a support material with a coating of bioactive glass that is deposited by means of pulsed laser ablation.

L8 ANSWER 3 OF 3 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER: 2000:58304 EPFULL

ENTRY DATE PUBLICATION: 20050824
UPDATE DATE PUBLICAT.: 20060803
DATA UPDATE DATE: 20060802
DATA UPDATE WEEK: 200631

TITLE (ENGLISH): SILVER-CONTAINING, SOL-GEL DERIVED BIOGLASS

COMPOSITIONS

TITLE (FRENCH): COMPOSITIONS DE VERRE BIOACTIF CONTENANT DE L'ARGENT ET

OBTENU PAR UNE TECHNIQUE SOL-GEL

TITLE (GERMAN): SILBERHALTIGE BIOGLAS-ZUSAMMENSETZUNGEN, DIE VON

SOL-GEL ZUSTAENDEN ABGELEITET WERDEN

INVENTOR(S): BELLANTONE, Maria, 17 Cowley Street, London SW1F 3LZ,

GB; COLEMAN, Nichola J., 17 Purbeck Road, Chatham, Kent

ME4 6ED, GB; HENCH, Larry, L., 38 Clarence Gate

Gardens, London NW1 6BA, GB

PATENT APPLICANT(S): Imperial College Innovations, 47 Prince's Gate, London

SW7 2QA, GB

PATENT APPL. NUMBER: 3103210 Clyde-Watson, Zoee, D Young & Co 120 Holborn, London AGENT: EC1N 2DY, GB AGENT NUMBER: 126711 DOCUMENT TYPE: Patent LANGUAGE OF FILING: English English LANGUAGE OF PUBL.: English LANGUAGE OF PROCEDURE: LANGUAGE OF TITLE: German; English; French PATENT INFO TYPE: EPB1 Granted patent PATENT INFORMATION: PATENT INFORMATION: NUMBER KIND NUMBER KIND DATE EP 1196150 B1 20050824 \_\_\_\_\_ WO 2000076486 20001221 DESIGNATED STATES: AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU MC NL PT APPLICATION INFO.: EP 2000-939832 A 20000614 WO 2000-US16207 A 20000614 PRIORITY INFO.: US 1999-139014P P 19990614 CITED NON PATENT LIT.: PATENT ABSTRACTS OF JAPAN vol. 017, no. 659 (C-1137), 7 December 1993 (1993-12-07) & JP 05 213621 A (TOKUYAMA SODA CO LTD), 24 August 1993 (1993-08-24); PATENT ABSTRACTS OF JAPAN vol. 1997, no. 08, 29 August 1997 (1997-08-29) & JP 09 110463 A (ION KOGAKU SHINKO ZAIDAN), 28 April 1997 (1997-04-28); JORUNAL OF BIOMEDICAL RESEARCH, vol. 51, no. 3, 5 September 2000 (2000-09-05), pages 484-490, CITED PATENT LIT .: WO 8501210 Α WO 9404657 Α WO 9907777 Α US 5071674 Α US 5126141 Α US 5298260 Α

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US 5874101

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NEWS			03	
NEWS				CA/CAplus F-Term thesaurus enhanced
NEWS			10	
	-			8.01c now available
NEWS	8	NOV	20	CAS Registry Number crossover limit increased to 300,000 in additional databases
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NEWS				CAS REGISTRY chemical nomenclature enhanced
NEWS	12			WPIDS/WPINDEX/WPIX manual codes updated
				GBFULL and FRFULL enhanced with IPC 8 features and
				functionality
NEWS	14	DEC	18	
				with preparation role
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NEWS		DEC		MARPAT to CA/Caplus accession number crossover limit increased
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NEWS	18			CA/CAplus enhanced with more pre-1907 records
NEWS	19	JAN	80	CHEMLIST enhanced with New Zealand Inventory of Chemicals
NEWS	EXP	RESS		VEMBER 10 CURRENT WINDOWS VERSION IS V8.01c, CURRENT
				CINTOSH VERSION IS V6.0c(ENG) AND V6.0Jc(JP),
			AN:	D CURRENT DISCOVER FILE IS DATED 25 SEPTEMBER 2006.
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=> s L1 and ((calcium oxide) or CaO)
         31755 L1 AND ((CALCIUM OXIDE) OR CAO)
=> s 12 and (silica or SiO2)
        17778 L2 AND (SILICA OR SIO2)
=> s L3 and ((boron oxide) or B2O3)
          3885 L3 AND ((BORON OXIDE) OR B2O3)
=> s 14 and ((magnesium oxide) or MgO)
          2589 L4 AND ((MAGNESIUM OXIDE) OR MGO)
=> s 15 and ((calcium floride) or CaF2)
           111 L5 AND ((CALCIUM FLORIDE) OR CAF2)
=> s L6 and ((phosphorus pentoxide) or P2O5)
            58 L6 AND ((PHOSPHORUS PENTOXIDE) OR P2O5)
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    ANSWER 1 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN
L7
ACCESSION NUMBER:
                         2006:154916 CAPLUS
DOCUMENT NUMBER:
                         145:319089
TITLE:
                         Synthesis of glass-ceramics in the CaO-
                         MgO-SiO2 system with B2O3,
                         P2O5, Na2O and CaF2 additives
AUTHOR (S):
                         Tulyaganov, D. U.; Agathopoulos, S.; Ventura, J. M.;
                         Karakassides, M. A.; Fabrichnaya, O.; Ferreira, J. M.
CORPORATE SOURCE:
                         Department of Ceramics and Glass Engineering,
                         University of Aveiro, Aveiro, 3810-193, Port.
SOURCE:
                         Journal of the European Ceramic Society (2006), 26(8),
                         1463-1471
                         CODEN: JECSER; ISSN: 0955-2219
```

Elsevier B.V.

PUBLISHER:

DOCUMENT TYPE: Journal LANGUAGE: English

AB Glass-ceramics based on the CaO-MgO-SiO2

system with limited amount of additives (B2O3, P2O5,

Na2O and CaF2) were prepared All the investigated compns. were melted at 1400°C for 1 h and quenched in air or water to obtain

transparent bulk or frit glass, resp. Raman spectroscopy revealed that the main constituents of the glass network are the silicates Q1 and Q2

units. SEM anal. confirmed liquid-liquid phase separation and that the glasses are

prone to surface crystallization Glass-ceramics were produced via sintering and

crystallization of glass-powder compacts made of milled glass-frit (mean particle  $% \left( 1\right) =\left( 1\right) +\left( 1\right) +$ 

size 11-15 μm). Densification started at 620-625°C and was almost complete at 700°C. Crystallization occurred at temps.

>700°C. Highly dense and crystalline materials, predominantly composed of diopside and wollastonite together with small amts. of akermanite and residual glassy phase, were obtained after heat treatment at 750°C

and 800°C. The glass-ceramics prepared at 800°C exhibited

bending strength of 116-141 MPa, Vickers microhardness of 4.53-4.65 GPa and thermal expansion coefficient ( $100-500^{\circ}C$ ) of 9.4-10.8+10-6

K-1. REFERENCE COUNT:

45 THERE ARE 45 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 2 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

2005:1291244 CAPLUS

DOCUMENT NUMBER:

144:260619

TITLE:

Formation of hydroxyapatite onto glasses of the

CaO-MgO-SiO2 system with B2O3, Na2O, CaF2 and P2O5

additives

AUTHOR(S): Agathopoulos, S.; Tulyaganov, D. U.; Ventura, J. M.

G.; Kannan, S.; Karakassides, M. A.; Ferreira, J. M.

F.

CORPORATE SOURCE:

Department of Ceramics and Glass Engineering, CICECO,

University of Aveiro, Aveiro, 3810-193, Port.

SOURCE:

Biomaterials (2006), 27(9), 1832-1840

CODEN: BIMADU; ISSN: 0142-9612

PUBLISHER:

Elsevier Ltd.

DOCUMENT TYPE:

Journal

LANGUAGE:

English

AB New bioactive glasses with compns. based on the CaO-MgO -SiO2 system and additives of B2O3, P2O5,

Na2O, and CaF2 were prepared The in vitro mineralization behavior was tested by immersion of powders or bulk glasses in simulated body fluid (SBF). Monitoring of ionic concns. in SBF and SEM observations at the surface of the glasses were conducted over immersion time. Raman and IR spectroscopy shed light on the structural evolution occurring at the

surface of the glasses that leads to formation of hydroxyapatite.

REFERENCE COUNT:

57 THERE ARE 57 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 3 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

2004:698152 CAPLUS

DOCUMENT NUMBER:

141:218962

TITLE:

Bioactive material for use in stimulating

vascularization

INVENTOR(S):

Day, Richard Michael

PATENT ASSIGNEE(S):

The North West London Hospitals Nhs Trust, UK

SOURCE:

PCT Int. Appl., 72 pp.

DOCUMENT TYPE:

CODEN: PIXXD2 Patent

LANGUAGE:

English

$D\Delta \Psi F N \Psi$	INFORMATION:

	PATENT NO.					KIND DATE			APPLICATION NO.						D.	DATE			
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				GE,	GH,	GM,	HR,	HU,	ID,	IL,	IN,	IS,	, JP,	KE,	KG,	ΚP,	KR,	KZ,	LC,
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				MC,	NL,	PT,	RO,	SE,	SI,	SK,	TR,	BF	, BJ,	CF,	CG,	CI,	CM,	GA,	GN,
									SN,										
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AB The present invention relates to a bioactive material, particularly one which comprises SiO2 and CaO and optionally Na2O and/or P2O5, for use in stimulating vascularization and pharmaceutical compns., wound dressings, tissue constructs and delivery systems which include such a bioactive material.

L7 ANSWER 4 QF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2004:206027 CAPLUS

DOCUMENT NUMBER: 141:179498

TITLE: On the microstructure of biocomposites sintered from

Ti, HA and bioactive glass

AUTHOR(S): Ning, C. Q.; Zhou, Y.

CORPORATE SOURCE: Center for Biomedical Engineering, Wenner Gren

Research Laboratory, University of Kentucky,

Lexington, KY, 40506, USA

SOURCE: Biomaterials (2004), 25(17), 3379-3387

CODEN: BIMADU; ISSN: 0142-9612

PUBLISHER: Elsevier Science Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

AB Sintering reactions and fine structures of the biocomposites prepared from

powder mixts. of titanium  $(\alpha\text{-Ti})$ , hydroxyapatite (HA) and

bioactive glass (BG) (SiO2-CaO-P2O5-B2O3-MgO-TiO2-CaF2) were

investigated by x-ray diffraction and TEM. The results showed that complex reactions among the starting materials mainly depended on the initial  ${\rm Ti/HA}$  ratios as well as the sintering temps. And the reaction could be expressed by the following illustrative equation:

Ti+Ca10(PO4)6(OH)2 CaTiO3+CaO+TixPy+(Ti2O)+(Ca4P2O9)+H2O.

REFERENCE COUNT: 26 THERE ARE 26 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 5 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2004:182250 CAPLUS

DOCUMENT NUMBER: 140:205207

TITLE: Biodegradable and bioactive glass

-ceramics, and method for fabricating the same

INVENTOR(S): Yu, Hyun Seung; Hong, Kug Sun; Kim, Hwan; Lee, Dong

Ho; Lee, Choon Ki; Chang, Bong Soon; Kim, Deug Joong;

Seo, Jun Hyuk; Lee, Jae Hyup; Park, Ki Soo

PATENT ASSIGNEE(S):

S. Korea

SOURCE: U.S. Pat. Appl. Publ., 17 pp.

CODEN: USXXCO

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

	PATENT NO.	KIND	DATE	APPLICATION NO.		DATE
	US 2004043053	A1		US 2003-648217		20030827
	KR 2004021111	Α	20040310	KR 2002-52566		20020902
PRIO	RITY APPLN. INFO.:			KR 2002-52566	Α	20020902
AB	Disclosed herein is	a biod	legradable an	nd bioactive glass		
	-ceramic fabricated	by mix	ing a slowly	biodegradable gla	ss-	
	ceramic and a highl	y biode	gradable gla	ss-ceramic in a		
	predetd. mixing rat	io wher	ein the bioa	activity is maintain	ned t	o be constant,
	and the biodegrdn.	rate is	controlled	by the mixing ratio	о. Т	he
	biodegradable and b	ioactiv	e glass-cera	amic is		
	fabricated from a c		-		e (	
	CaO), silica (SiO2)	_				
	oxide (B2O3), magne	sium ox	ide (			
	MgO), calcium fluor			sphorus		
	pentoxide (P2O5).					,
	bioactive glass-cer					
	method is disclosed		_		hvdro	xvcarbonated
	apatite layers were				_	-
	out 1 day after soa					
	specimens were high					
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L7 ANSWER 6 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

2003:117764 CAPLUS

DOCUMENT NUMBER:

138:157658

TITLE:

Alumina-based glass-ceramics with high hardness for

use as abrasives

INVENTOR (S):

Rosenflanz, Anatoly Z.

PATENT ASSIGNEE(S):

3M Innovative Properties Company, USA

demonstrated that the biodegrdn. of the specimens had actively proceeded.

PCT Int. Appl., 112 pp. SOURCE: CODEN: PIXXD2

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT: 18

PATENT INFORMATION:

PATENT NO.				KIND DATE		APPLICATION NO.					DATE							
WO 2003011784 WO 2003011784															20020802			
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		CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	ES,	FI,	GB,	GD,	GE,	GH,	
		GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	ΚE,	KG,	ΚP,	KR,	ΚZ,	LC,	LK,	LR,	
		LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,	NO,	NZ,	OM,	PH,	
		PL,	PT,	RO,	RU,	SD,	SE,	SG,	SI,	SK,	SL,	ТJ,	TM,	TN,	TR,	TT,	TZ,	
		UA,	UG,	UŻ,	VN,	YU,	ZA,	ZM,	zw									
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		CH,	CY,	CZ,	DE,	DK,	EE,	ES,	FI,	FR,	GB,	GR,	ΙE,	IT,	LU,	MC,	NL,	
		PT,	SE,	SK,	TR,	BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,	
		ΝE,	SN,	TD,	TG													
CA	2454	646			A1		2003	0213	(	CA 20	002-2	2454	546		2	0020	302	
CA	2455	902			A1		2003:	1218	(	CA 20	002-	2455	902		2	0020	302	
ΑU	2002	3679:	31		A1	;	2003:	1222	i	AU 20	002-3	3679:	31		2	0020	302	
EΡ	1432	659			A1		2004	0630	1	EP 20	002-	7504:	13		2	0020	302	
	R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	IT,	LI,	LU,	NL,	SE,	MC,	PT,	
	•	ΙE,	SI,	LT,	LV,	FI,	RO,	MK,	CY,	AL,	TR,	BG,	CZ,	EE,	SK			
ΕP	1432	660			A1		2004	0630	1	EP 20	002-	8073	69		2	0020	302	
	R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	IT,	LI,	LU,	NL,	SE,	MC,	PT,	

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IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, SK
     BR 2002011558
                          Α
                                 20040713
                                             BR 2002-11558
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                                             CN 2002-814994
     CN 1537082
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                                 20041013
     CN 1537084
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                                             CN 2002-815119
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     JP 2004536769
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                                 20041209
                                             JP 2003-516981
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     CN 1558876
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                                 20041229
                                             CN 2002-818937
                                                                     20020802
     CN 1582262
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                                 20050216
                                             CN 2002-815127
     JP 2005519846
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     CN 1649802
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     BR 2002011578
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                                 20060404
                                             BR 2002-11578
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PRIORITY APPLN. INFO.:
                                             US 2001-922526
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                                             US 2001-922527
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                                             US 2001-922528
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                                             US 2001-922530
                                                                  Α
                                             WO 2002-US24657
                                                                     20020802
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                                             WO 2002-US24658
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                                                                    20020802
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AB Glass-ceramics for use as abrasive particles with high hardness contain alumina and other oxides with less than 10 weight% in As2O3, B2O3, GeO2, P2O5, SiO2, TeO2, and V2O5 combined and are formed by heat-treatment of the glass compns. into glass-ceramics. The glass-ceramics also contain other components than alumina (such as La2O3, ZrO2, Al, Y2O3, Gd2O3, Mg, MgO, TiO2, CaF2, Nb2O5, Ta2O5, SrO, Mn2O3, Cr2O3 and/or CeO2) and may be crushed to form the abrasive particles. The abrasive particles can be incorporated into a variety of abrasive articles such as bonded abrasives, coated abrasives, nonwoven abrasives and abrasive brushes.

L7 ANSWER 7 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

2003:117762 CAPLUS

DOCUMENT NUMBER:

138:157584

TITLE:

Alumina-zirconia-based glass-ceramics with high

hardness for use as abrasives

INVENTOR(S):

Rosenflanz, Anatoly Z.; Celikkaya, Ahmet; Anderson,

Thomas J.

PATENT ASSIGNEE(S):

3M Innovative Properties Company, USA

SOURCE:

PCT Int. Appl., 104 pp.

DOCUMENTO MILI

CODEN: PIXXD2

DOCUMENT TYPE: LANGUAGE: Patent English

18

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.					KIND DATE			APPLICATION NO.						DATE					
WO	2003	0117	82		A2	_	2003	0213							20020802				
																CH,			
		CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	ES,	FI,	GB,	GD,	GE,	GH,		
		GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	ΚP,	KR,	ΚZ,	LC,	LK,	LR,		
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	RW:	•		-													BG,		
						•	•									MC,			
		•	•	•	•	•	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,		
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	2002																		
EP	1430																		
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EP	1432																		
	R:	•	•	•	•	•	•	•		•	•	•	•	-		MC,	ΡΙ,		
CNI	1537							MK,								0020	8 N 2		
	1537																		
CIV	100/	004			A		2004	TOTO	,	C14 Z	002-	0131	1)		2	0020	002		

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                                           CN 2002-818937
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    CN 1558876
                         Α
                               20041229
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                                                                  20020802
    CN 1582262
                               20050216
                                           CN 2002-815127
                                                                  20020802
                               20050707
                                           JP 2004-511231
    JP 2005519846
                                           CN 2002-819113
                               20050803
                                                                  20020802
    CN 1649802
    BR 2002011578
                               20060404
                                           BR 2002-11578
                                                                  20020802
                                                               A 20010802
                                           US 2001-922526
PRIORITY APPLN. INFO.:
                                                               A 20010802
                                           US 2001-922527
                                           US 2001-922528
                                                                  20010802
                                                               Α
                                           US-2001-922530
                                                                  20010802
                                                               Α
                                           WO 2002-US24658
                                                               W
                                                                  20020802
                                                               W 20020802
                                           WO 2002-US24714
    Glass-ceramics for use as abrasive particles with high hardness contain
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AB Glass-ceramics for use as abrasive particles with high hardness contain alumina, stabilized zirconia and other oxides with less than 20 weight% in As203, B203, Ge02, P205, Si02, Te02, and V205 combined and are formed by heat-treatment of the glass compns. into glass-ceramics. The glass-ceramics also contain other oxide components than alumina and zirconia such as La203, Al, Y203, Gd203, Mg, Mg0, Ti02, CaF2, Nb205, Ta205, Sr0, Mn203, Cr203 and/or Ce02 and may be crushed to form the abrasive particles. The abrasive particles can be incorporated into a variety of abrasive articles such as bonded abrasives, coated abrasives, nonwoven abrasives and abrasive brushes.

L7 ANSWER 8 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2002:868867 CAPLUS

DOCUMENT NUMBER: 137:356884

TITLE: Bioactive calcium silicate glass with antimicrobial

effects as additive for polymers

INVENTOR(S): Fechner, Joerg Hinrich; Zimmer, Jose; Schnabel,

Roland; Schnell, Rupert

PATENT ASSIGNEE(S): Schott Glas, Germany; Carl-Zeiss-Stiftung

SOURCE: PCT Int. Appl., 16 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: German

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

•	PATENT NO.					KIND DATE			APPLICATION NO.						DATE			
	WO 2002090278					A1 20021114			WO 2002-EP4991						20020507			
		W:	ΑE,	AG,	AL,	AM,	AT,	AU,	ΑZ,	BA,	BB,	BG,	BR,	BY,	BZ,	CA,	CH,	CN,
			CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	ES,	FI,	GB,	GD,	GE,	GH,
			GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	KP,	KR,	KZ,	LC,	LK,	LR,
			LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,	NO,	ΝŻ,	OM,	PH,
			PL,	PT,	RO,	RU,	SD,	SE,	SG,	SI,	SK,	SL,	ТJ,	TM,	TN,	TR,	TT,	TZ,
			UA,	UG,	US,	UZ,	VN,	YU,	ZA,	ZM,	ZW,	AM,	ΑZ,	BY,	KG,	ΚZ,	MD,	RU,
			TJ,	TM														
		RW:	GH,	GM,	KE,	LS,	MW,	MZ,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AT,	BE,	CH,
			CY,	DE,	DK,	ES,	FI,	FR,	GB,	GR,	ΙE,	IT,	LU,	MC,	NL,	PT,	SE,	TR,
			BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,	ΝE,	SN,	TD,	ТG
	DE	1012	2262			A1		2002	1121	1	DE 2	001-	1012	2262		2	0010	508
	JP	2004	5266	57		$\mathbf{T}$		2004	0902	,	JP 2	002-	5873	64		2	0020	507
	US	2004	1370	75		A1		2004	0715	1	US 2	003-	4770	98		2	0031	107
PRIO	RIT	Y APP	LN.	INFO	. :						DE 2	001-	1012	2262	7	A 2	0010	508
										1	WO 2	002-	EP49	91	1	w 2	0020	507
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AB The bioactive glass is used as additive for polymers, whereby the bioactive glass contains SiO2
40-90, CaO 4-45, Na2O 0-35, P2O5 2-16, CaF2
0-25, B2O3 0-10, K2O 0-8, and/or MgO 0-5 weight%. The invention also relates to a polymer as supporting material comprising. 1-30 weight% bioactive glass in relation to its overall weight
The bioactive glass can be added with Ag+, Cu+, Cu2+, and/or Zn+. The bioactive glass as an antimicrobial, and fumigant additive for polymers is suitable in households, packaging,

food processing, sealing materials, clothing, medical materials, sanitary materials, automotive industry, building industry, as well as plastic coatings, or adhesives.

REFERENCE COUNT: 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 9 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2002:658759 CAPLUS

DOCUMENT NUMBER: 137:206591

TITLE: Dental/medical compositions comprising degradable

polymers

Patent

INVENTOR(S):
Jia, Weitao; Jin, Shuhua

PATENT ASSIGNEE(S): Pentron Corp., USA

SOURCE: U.S. Pat. Appl. Publ., 10 pp., Cont.-in-part of U.S.

Ser. No. 638,206.

CODEN: USXXCO

DOCUMENT TYPE:

LANGUAGE: English

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 2002120033	A1	20020829	US 2001-5298	20011205
US 6787584 US 6455608	B2 B1	20040907 20020924	US 2000-638206	20000811
WO 2002078646 RW: AT, BE, CH,	A1 CY, DE	20021010 , DK, ES,	WO 2001-US46526 FI, FR, GB, GR, IE,	20011205 IT, LU, MC, NL,
PT, SE, TR US 2004249015	A1	20041209	US 2004-862177	20040604
PRIORITY APPLN. INFO.:	• • • •	20012203	US 2000-638206 US 2000-251408P	A2 20000811 P 20001205
•			US 1999-148887P	P 19990813

Curable compns. comprising degradable macromonomers having one or more AB terminal acrylate or methacrylate functionality, a curing composition, a filler composition comprising bioactive particles of bioactive glass , bioactive glass-ceramics, calcium phosphates, calcium apatites, or mixts. thereof and optionally one or more polymerizable acrylate or methacrylate comonomers. Degradable macromonomers are manufactured from cyclic monomers with compds. having acrylate or methacrylate functionality. Depending on their use, the degradable macromonomer compns. further comprise one or more organic or inorg. fillers, including a calcium-based compound and/or a radiopacity-imparting agent. The compns. are particularly suitable for root canal sealants, implants, bone cements, and pulp capping materials. The degradation of cured PLAMA (a polylactide-HEMA)/TEGDMA for varying quantities of PLAMA was studied in a buffer solution of pH 7 over time. curing of PLAMA/TEGDMA was performed for 2 min by using a curing composition comprising 0.2% by weight camphorquinone and 0.2% DEAEMA. PLAMA and TEGDMA were mixed to give various weight percents (based on the total resin composition)

US 2001-5298

A3 20011205

of PLAMA. The degradation of the samples is related to the weight percent of PLAMA used to prepare the samples, such that degradation of the samples increases with the increasing quantities of PLAMA relative to TEGMA. When no PLAMA is present, no degradation is observed

REFERENCE COUNT: 50 THERE ARE 50 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 10 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2002:367154 CAPLUS

DOCUMENT NUMBER: 136:358501

TITLE: Dry press-molding and slip-casting preparation of

magnesium aluminosilicate glass-ceramic

pressure sensor support bases

Xue, Liang A.; Bernot, Anthony J.; Hughes, Grenville; INVENTOR(S):

Lindberg, Laura

PATENT ASSIGNEE(S):

AlliedSignal, Inc., USA

SOURCE: U.S., 8 pp., Cont.-in-part of U.S. 6,058,780.

CODEN: USXXAM

DOCUMENT TYPE:

LANGUAGE:

Patent English

FAMILY ACC. NUM. COUNT:

2

PATENT INFORMATION:

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KIND
                               DATE APPLICATION NO.
    PATENT NO.
    _____
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    US 6387318 B1
WO 2001010788 A1
                               20020514 US 1999-371676 19990810
20010215 WO 2000-US21866 20000810
           AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE,
            DK, EE, ES, FI, GB, GE, GH, GM, HU, ID, IL, IN, IS, JP, KE, KG,
            KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX,
            NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT,
            UA, UG, UZ, VN, YU, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
        RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY,
            DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ,
             CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
                                           US 1997-986253 A2 19971205
US 1999-371676 A 19990810
PRIORITY APPLN. INFO.:
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Dry-pressing and slip-casting processes for the manufacture of aluminosilicate AB glass-ceramic components used in sensors are described. The starting powder compns. consist of SiO2 (42-59 weight%), Al2O3 (17-35 weight%), MgO (2-25 weight%) and also BaO (0-20 weight%), TiO2 (0-12 weight%), ZnO (0-10 weight%), CaO (0-10 weight%), B2O3 (0-5 weight%), P2O5 (0-5 weight%), ZrO2 (0-5 weight%), alkali oxides such as K2O or Na2O (0-3 weight%) and 0-1 weight% of additives such as Fe2O3, V2O5 and/or As203. Some of the ingredients can also be present in the form of fluoride such as MgF2, BaF2 or CaF2 up to 10 weight%. Addnl. ceramic additives up to 30 weight%, which are as second phase (e.g., Al203, Y203 doped Zr02, Si02, mullite, zircon, carbides and/or nitrides), are not part of the glass composition, but are introduced to enhance both the mech. strength and toughness and to modify the coefficient of thermal expansion. In the dry pressing process the powder is milled, blended with a binder and then subjected to a drying/granulation process. The powder is inserted in a die and pressed for a period of time until the compact is removed and then sintered to produce ceramic covers or headers. In the slip casting process, the powder is milled and dispersed in a water. The slip is then poured or pumped into a permeable gypsum casting mold. When either the liquid has been completely sucked away by the mold or the desired depositing thickness has been reached, the cast green ceramic part is removed from the mold, dried and then sintered into a ceramic support base.

REFERENCE COUNT:

63 THERE ARE 63 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 11 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2001:272488 CAPLUS

DOCUMENT NUMBER: 134:370435

TITLE: Crystallization behavior of some selected bioglasses

AUTHOR(S): El-Batal, H. A.; Azooz, M. A.; El-Batal, F. H.

CORPORATE SOURCE: Glass Research Department, National Research Centre,

Cairo, Egypt

Applied Mineralogy: In Research, Economy, Technology, SOURCE:

Ecology and Culture, Proceedings of the International Congress on Applied Mineralogy, 6th, Goettingen, Germany, July 17-19, 2000 (2000), Volume 1, 123-125.

Editor(s): Rammlmair, Dieter. A. A. Balkema:

Rotterdam, Neth.

CODEN: 69BDPW

DOCUMENT TYPE:

Conference

English LANGUAGE:

Crystallization behavior in some bioglasses of the system Na2O-CaO-

SiO2-P2O5 together with the introduction of B2O3 , MgO, K2O and CaF2 were studied. Nucleation and

crystallization regimes were carried out by parameter obtained from DTA and dilatometric measurements. The crystalline phases identified using x-ray diffraction studies and the microstructure was examined by scanning electron

microscope.

THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 12 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

ANSWER 12 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2001:196240 CAPLUS

DOCUMENT NUMBER: 135:111922

Chemical durability of commercial silicate glasses. I. TITLE:

Material characterization

AUTHOR (S): Jedlicka, A. B.; Clare, A. G.

New York State College of Ceramics, Alfred University, CORPORATE SOURCE:

Alfred, NY, 14802, USA

Journal of Non-Crystalline Solids (2001), 281(1-3), SOURCE:

6-24

CODEN: JNCSBJ; ISSN: 0022-3093

PUBLISHER: Elsevier Science B.V.

DOCUMENT TYPE: Journal English LANGUAGE:

Six com. silicate glasses; silica, sodalime silicate, two AB fiberglass compns. and two Bioglass compns. were subjected to

three solns.: distilled water, Dulbecco's phosphate buffered saline solution

and

Ham's F-12 1x cell culture media under the exact exptl. conditions that would be encountered during a cell culture study. For companion, a binary sodium silicate glass was also exposed. Weight loss, diffuse reflectance IR spectroscopy (DRIFTS) and potentiometric titration were used to determine the chemical evolution of the substrates during a typical period for cell culturing. The silica, sodalime silicate and high-

silica fiberglass material showed only small changes in all cases except for differences in OH active site concentration The Bioglass compns. and the low-silica fiberglass exhibited solution-dependent dynamic surface chemical The sodium silicate was too dynamic for even the most aggressive buffering system. The purpose of this study was to

elucidate cell behavior to be reported in a later paper.

THERE ARE 17 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 17 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

ANSWER 13 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

2001:50441 CAPLUS

DOCUMENT NUMBER:

134:90878

TITLE:

Usage of bioactive glass as

preservative for cosmetic and pharmaceutical

preparations

INVENTOR(S):

Kessler, Susanne; Lee, Sean

PATENT ASSIGNEE(S):

Schott Glas, Germany PCT Int. Appl., 13 pp.

SOURCE: CODEN: PIXXD2

DOCUMENT TYPE:

Patent

LANGUAGE:

German

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2001003650	A2	20010118	WO 2000-DE2231	20000707
WO 2001003650	A3	20010927		

2001003650 A3 20010927 W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR,

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HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT,
             LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU,
             SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN,
             YU, ZA, ZW
        RW: GH, GM, KE, LS, MW, MZ, SD, SL; SZ, TZ, UG, ZW, AT, BE, CH, CY,
             DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ,
             CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
                          A1
                                20010118
                                          · CA 2000-2374395
                                                                    20000707
    BR 2000012330
                          Α
                                20020319
                                            BR 2000-12330
                                                                    20000707
    EP 1194113
                                20020410
                                            EP 2000-956075
                                                                    20000707
                          A2
            AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO
    TR 200103723
                         т2
                                20020621
                                            TR 2001-3723
                                                                    20000707
    HU 200201821
                          A2
                                20021128
                                            HU 2002-1821
                                                                    20000707
    JP 2003504317
                          т
                                20030204
                                            JP 2001-508934
                                                                    20000707
    NZ 516136
                                20040227
                                            NZ 2000-516136
                                                                    20000707
                          Α
    AU 780131
                          B2
                                20050303
                                            AU 2000-68184
                                                                    20000707
    NO 2002000082
                          Α
                                20020108
                                            NO 2002-82
                                                                    20020108
     ZA 2002000155
                          Α
                                20040211
                                            ZA 2002-155
                                                                    20020108
PRIORITY APPLN. INFO.:
                                            DE 1999-19932239
                                                                 A 19990709
                                            WO 2000-DE2231
                                                                 W 20000707
```

AB The invention relates to a preservative which contains a bio-active glass and a protic solvent. The inventive preservative is used preferably for preserving cosmetic and pharmaceutical prepns., in particular for creams, lotions, lipsticks, make-up compns. and/or tinctures. The bioactive glass contains in weight/weight%: SiO2 40-60; CaO 10-30; Na2O 10-35; P2O5 2-8; CaF2 0-10; B2O5 0-8; K2O or MgO 0-5.

L7 ANSWER 14 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 2000:718225 CAPLUS

DOCUMENT NUMBER: 133:300023

TITLE: Manufacture of strong miserite glass ceramics for bone

implant materials

INVENTOR(S): Andrus, Ronald L.; Beall, George H.; Pinckney, Linda

R.; Wolcott, Christine C.

PATENT ASSIGNEE(S): Corning Incorporated, USA

SOURCE: U.S., 7 pp.
CODEN: USXXAM

DOCUMENT TYPE: Patent

LANGUAGE: English FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 6130178	A	20001010	US 1999-286158	19990405
PRIORITY APPLN. INFO.:			US 1998-82127P P	19980416

AB The glass ceramics having high strength and toughness and containing miserite as a predominant crystal phase are described. The glass ceramics comprise SiO2 40-68, CaO 12-35, CaF2 8-20, K2O 4-10.5,

Al2O3 0-5, B2O3 0-5, P2O5 0-15, R2O3 0-4 weight%, wherein

R represents Y3+ and rare earth metals in the lanthanide series, and 0-5 weight% of optional constituents selected from MgO, SrO, BaO, Na2O,

Nb205, ZrO2, and ZnO, and 0-10 weight% of optional constituents selected from Nb205 and TiO2, and 0-2 weight% Li2O as an optional constituent. The glass ceramics may contain secondary phases of cristobalite (SiO2),

fluorite (CaF2), xonotlite (Ca6Si6O17F2), and fluorapatite

(Ca5(PO4)3F), and small amts. of other calcium silicate phases such as wollastonite (CaSiO3).

REFERENCE COUNT: 13 THERE ARE 13 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 15 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN ACCESSION NUMBER: 2000:285703 CAPLUS

DOCUMENT NUMBER:

132:283065

TITLE:

Bioactive glass composition and methods of treating sick tooth

INVENTOR(S):

Litkowski, Leonard J.; Hack, Gary D.; Greenspan, David

PATENT ASSIGNEE(S):

Univ. of Maryland at Baltimore, USA; Usbiomaterials

SOURCE:

Faming Zhuanli Shenqing Gongkai Shuomingshu, 25 pp.

CODEN: CNXXEV

DOCUMENT TYPE:

Patent

LANGUAGE:

Chinese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.		DATE
				-	
CN 1213355	Α	19990407	CN 1997-193085		19970129
CN 1103750	В	20030326			
ни 9901760	A2	19991028	HU 1999-1760		19970129 -
PT 877716	${f T}$	20050228	PT 1997-906489		19970129
ES 2230597	Т3	20050501	ES 1997-906489		19970129
US 5735942	A	19980407	US 1997-895329	•	19970716
US 6338751	В1	20020115	US 1998-52647		19980331
PRIORITY APPLN. INFO.:			US 1996-10795P	P	19960129
			US 1996-597936	Α	19960207
			US 1997-789909	Α	19970129
			US 1997-895329	A1	19970716

The glass composition having bioactivity and biocompatibility is composed of: AΒ SiO2 40-60, CaO 10-30, Na2O 10-35, P2O5 2-8, CaF2 0-25, B2O3 0-10, K2O 0-8, and MgO 0-5 weight%, where the glass composition has glass particles <90 µm and mineral additive particles <10 µm. The composition is mixed with other substance, and used for prevention and treatment of early dental caries, and for repair of affected tooth. The other substance is tooth paste, lining, substrate, gel, repair material, glycerol gel, preventive, and/or indirect pulp-cap, etc. The methods of treating sick or affected tooth using the title composition are also claimed.

ANSWER 16 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN L7

ACCESSION NUMBER:

2000:205201 CAPLUS

DOCUMENT NUMBER:

132:211608

TITLE:

Preparation of luminous enamel glazes and articles

INVENTOR(S):

Zhang, Yujun

PATENT ASSIGNEE(S):

Lunbo Commercial and Industrial Group Co., Ltd., Peop.

Rep. China

SOURCE:

Faming Zhuanli Shenqing Gongkai Shuomingshu, 6 pp.

CODEN: CNXXEV

DOCUMENT TYPE:

Patent

LANGUAGE:

Chinese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
CN 1198413	Α	19981111	CN 1998-110267	19980605
CN 1058947	В	20001129		
PRIORITY APPLN. INFO.:			CN 1998-110267	19980605
AB The enamel glaze :	is compos	sed of lumir	nous powder 20-45, ba	se glaze 45-100,
			ıs powder is MO.nAl20	
Mg, Ba, and/or Ca	; n = 1-3	3). The add	ditive is CMC, clay,	and/or xanthan
gum. The base gla	aze is co	omposed of S	SiO2 10-70, Al2O3 2-1	5,
CaO 0-5, CaF2 0-8	, K2O 1-1	LO, Na2O 6-1	L5, ZnO 0-3, PbO	
0-50, P2O5 0-5, S:	ro 5-35,	MgO 0-6, Ba	aO 0-20,	
B203 6-32, and Li	20 0-6 pa	art. The pi	cocess comprises mixi	ng the

raw material for base glaze, sintering at 1,000-1.350°, annealing

with water, mixing with the other raw material, and grinding. The enamel product is prepared by coating the glaze on white overglaze, and baking at 780-820° for 1-3 min.

ANSWER 17 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1998:326314 CAPLUS

DOCUMENT NUMBER: 129:44046

TITLE: Surface nucleated crystallization of wollastonite and

wollastonite-apatite glasses

Ciecinska, M.; Stoch, L.; Duda, A. AUTHOR(S):

Univ. Mining Metallurgy, Krakow, 30-059, Pol. CORPORATE SOURCE: Prace Komisji Nauk Ceramicznych, Ceramika (Polska SOURCE:

Akademia Nauk) (1996), 49(Properties of Glasses with

Modifying Compounds), 109-123 CODEN: PKNCE6; ISSN: 0860-3340 Polskie Towarzystwo Ceramiczne

PUBLISHER: DOCUMENT TYPE: Journal English LANGUAGE:

The sintering/simultaneous crystallization of glasses of composition which enables

formation of wollastonite or wollastonite-apatite phases were studied.

Compns. studied include P2O5 14-CaO 47-SiO2

34-MgO 5-CaF2 0.5, P2O5 14-CaO 47-SiO2 34-MgO 5-B2O3 0.5, CaO 46-

SiO2 49-Na2O 5, and CaO 43.5-SiO2 46.5-Na2O 10

wt%. After melting, quenching, granulating, compacting, and

sintering/crystallizing, the glass-ceramic samples were examined by DSC and x-ray diffraction.

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

9

ANSWER 18 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN L7 ACCESSION NUMBER: 1997:525846 CAPLUS

DOCUMENT NUMBER: 127:195535

TITLE:

Bioactive glass compositions, and treatment method using the bioactive

glass

Patent

Litkowski, Leonard J.; Hack, Gary D.; Greenspan, David INVENTOR(S):

PATENT ASSIGNEE(S): University of Maryland, USA; Usbiomaterials Corp.;

Litkowski, Leonard J.; Hack, Gary D.; Greenspan, David

THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS

SOURCE: PCT Int. Appl., 34 pp.

CODEN: PIXXD2

DOCUMENT TYPE:

English LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

REFERENCE COUNT:

PA'	rent :	NO.			KINI	)	DATE			APPL	ICAT:	I NOI	. OV		D	ATE	
						_											
WO	9727	148			A1		1997	0731	1	WO 1	997-1	JS17	85		19	9970:	129
	W:	AL,	AM,	AT,	AU,	ΑZ,	BA,	BB,	BG,	BR,	BY,	CA,	CH,	CN,	CU,	CZ,	DE,
		DK,	EE,	ES,	FI,	GB,	GE,	ΗU,	IL,	IS,	JP,	ΚE,	KG,	ΚP,	KR,	ΚZ,	LC,
		LK,	LR,	LS,	LT,	LU,	LV,	MD,	MG,	MK,	MN,	MW,	MX,	NO,	ΝZ,	PL,	PT,
		RO,	RU,	SD,	SE,	SG,	SI,	SK,	TJ,	TM,	TR,	TT,	UA,	ŪĠ,	US,	UΖ,	VN
	RW:	KE,	LS,	MW,	SD,	SZ,	UG,	ΑT,	ΒE,	CH,	DE,	DK,	ES,	FI,	FR,	GB,	GR,
		IE,	ΙT,	LU,	MC,	NL,	PT,	SE,	BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	ML,
		MR,	ΝE,	SN,	TD,	TG											
BR	9707	219			Α		1999:	1228		BR 1:	997-	7219			19	9970:	126
CA	2244	722			A1		1997	0731		CA 1:	997-	2244	722		1.9	9970	129
CA	2244	722			С		2006	0829									
ΑU	9721	171			Α		1997	0820		AU 1	997-	2117:	1		19	99702	L29
ΑU	7236	59			в2		2000	0831									
ΕP	8777	16			A1		1998	1118		EP 1:	997-	9064	89		19	9970	129

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20041013
    EP 877716
                         В1
        R: AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LI, LU, MC, NL,
            PT, SE
    HU 9901760
                         A2
                               19991028
                                           ни 1999-1760
                                                                  19970129
    NZ 331514
                               20000327
                                           NZ 1997-331514
                                                                  19970129
                         Α
    JP 2001525779
                         \mathbf{T}
                                           JP 1997-527131
                                                                  19970129
                               20011211
                         \mathbf{T}
                                           AT 1997-906489
                                                                  19970129
    AT 279380
                               20041015
    PT 877716
                         \mathbf{T}
                               20050228
                                         PT 1997-906489
                                                                  19970129
                         T3
                                          ES 1997-906489
    ES 2230597
                               20050501
                                                                  19970129
                                          US 1997-895329
    US 5735942
                         Α
                               19980407
                                                                  19970716
                                           US 1998-52647
    US 6338751
                         B1
                               20020115
                                                                  19980331
                               19980923
                                           NO 1998-3490
    NO 9803490
                        Α
                                                                  19980729
    HK 1019222
                        A1
                               20030711
                                           HK 1999-104392
                                                                  19991007
                                                              P 19960129
PRIORITY APPLN. INFO.:
                                           US 1996-10795P
                                           US 1996-597936
                                                              A 19960207
                                           US 1997-789909
                                                              A 19970129
                                           WO 1997-US1785
                                                              W 19970129
                                                               A1 19970716
                                           US 1997-895329
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AB A novel silica-based bioactive glass composition is described that can be used in conjunction with a delivery agent such as a toothpaste, gel, etc., having a particle size range <90 µm which will form a rapid and continuous reaction with body fluids due to the immediate and long-term ionic release of Ca and P from the core silica particles, to produce a stable crystalline hydroxycarbonate apatite layer deposited onto and into the dentin tubules for the immediate and long-term reduction of dentin hypersensitivity and tooth surface remineralization. Thus a bioglass composition contained SiO2 45, CaO 24.5, Na2O 24.5, and P2O5 6% by weight The composition was placed on dentin slabs. The exposure times to the dentin varied between 2 min with scrubbing to 3 days. The formation of hydroxyapatite on the dentin surface was confirmed by IR spectroscopy.

L7 ANSWER 19 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1993:523416 CAPLUS

DOCUMENT NUMBER: 119:123416

TITLE: Glass for biocompatible glass ceramics

INVENTOR(S): Tulyaganov, Dilshat U.; Abdullaev, Sharif Yu.;

Makhkamov, Mokhir E.; Aripova, Mastura Kh.

PATENT ASSIGNEE(S): Tashkentskij g med institut, USSR

SOURCE: U.S.S.R. From: Izobreteniya 1992, (23), 92.

CODEN: URXXAF

DOCUMENT TYPE: Patent LANGUAGE: Russian

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

AUTHOR(S):

PATENT NO.	KIN	D DATE	APPLICATION NO.	DATE
SU 1742239	A1	19920623	SU 1990-4839542	19900618
PRIORITY APPLN. INFO.:			SU 1990-4839542	19900618
AB Crystallization rat	e is	increased and	articles of natural	color tone are
obtained				

when the glass contains MgO 0.89-15.2 and Fe2O3 0.15-0.25 and/or Mn2O3 0.05-15.02 in addition to SiO2 27.74-46.91, Al2O3 1.81-21.28, CaO 28.80-29.94, P2O5 5.21-14.312, B2O3 0.99-2.90, and CaF2 0.96-2.62 weight%.

L7 ANSWER 20 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1992:518454 CAPLUS

DOCUMENT NUMBER: 117:118454

TITLE: Influence of substituting boron trioxide for calcium

fluoride on the bonding behavior to bone of

glass-ceramics containing apatite and wollastonite Kitsugi, Toshiaki; Yamamuro, Takao; Nakamura, Takashi;

Yoshii, Satoru; Kokubo, Tadashi

CORPORATE SOURCE:

Fac. Med., Univ. Kyoto, Kyoto, Japan Biomaterials (1992), 13(6), 393-9

CODEN: BIMADU; ISSN: 0142-9612

DOCUMENT TYPE: LANGUAGE:

SOURCE:

Journal English

Glass-ceramics containing crystalline oxyfluoroapatite [Ca10(PO4)(O,F2)] and wollastonite (CaSiO3) (designated AWGC) are reported to have a fairly high mech. strength as well as the capability of forming a chemical bond with bone tissue. The chemical composition is MgO 4.6, CaO 44.9,

SiO2 34.2, P2O5 16.3, and CaF2 0.5 in weight

ratio. In this study the influence of substituting B2O3 for CaF2 on the bonding behavior of glass-ceramics containing apatite and wollastonite to bone tissue was investigated. Two kinds of glass-ceramics containing apatite and wollastonite were prepared CaF2 0.5 was replaced with B203 at 0.5 and 2.0 in weight ratio (designated AWGC-0.5B and AWGC-2.0B). Rectangular ceramic plates (15 + 10 + 2 mm, abraded with Number 2000 alumina powder) were

implanted into a rabbit tibia. The failure load, when an implant detached from the bone, or the bone itself broke, was measured. The failure load of AWGC-0.5B was 8.00 ± 1.82 kg at 10 wk after implantation and 8.16 ± 1.36 kg at 25 wk after implantation. The failure load of OSGC-2B was  $8.08 \pm 1.70$  kg at 10 wk after implantation and  $9.92 \pm 2.46$  kg at 25 wk after implantation. None of the loads for the two kinds of glass-ceramics decreased as time passed. Giemsa surface staining and contact microradiog. revealed direct bonding between glass-ceramics and SEM-EPMA showed a Ca-P rich layer (reaction zone) at the interface of ceramics and bone tissue. There was no difference of the reaction zone thickness between AWGC-0.5B and AWGC-2.0B. The substitution B2O3 for CaF2 did not influence the bonding ability of the glass-ceramics. Boron neither promotes the dissoln. of the glass-ceramics

nor influences the bond formation at the interface of ceramics and bone.

ANSWER 21 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

1986:191677 CAPLUS

DOCUMENT NUMBER:

104:191677

TITLE:

Glass-ceramics

INVENTOR(S): PATENT ASSIGNEE(S): Shibuya, Takehiro; Hashibe, Yoshio Nippon Electric Glass Co., Ltd., Japan

SOURCE:

Jpn. Kokai Tokkyo Koho, 4 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	JP 61017442 JP 04006659	 А В	19860125 19920206	JP 1984-139563	19840704
	RITY APPLN. INFO.:	Б	19920200	JP 1984-139563	19840704
				ight % substances which	are composed
	of SiO2 41-56, CaO		•		
				e consisting of minute	
	dispersed in a glas	s matri	x. The matr	ix preferably contains	≤10
	weight % of compds.	select	ed from BaO,	SrO, ZnO, B2O3, Al2O3,	TiO2,
	ZrO2, Nb2O3, Ta2O5,	and Ca	F2. The gla	ss powder (≤200	
	mesh) may also be c	ompacte	d, heated at	the sintering temperat	ure, and
				mperature The glass-ce not chip or crack durin	
it			,		<i>y</i>

has excellent mech. strength, elec. insulation, and dielec. characteristics. Thus, a raw material mixture composed of SiO2 41.8, P205 7.6, CaO 48.8, and MgO 1.8 weight % was melted at 1400-1500° in a Pt crucible for 4 h, allowed to flow between water-cooled rollers to give ribbon glass, and crushed to give powder ( $\leq$ 200 mesh). The powder was compacted, and sintered at 1050° and crystallized The glass-ceramic showed bending strength 1850 kg/cm2.

L7 ANSWER 22 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1985:459356 CAPLUS

DOCUMENT NUMBER: 103:59356

TITLE: Glass ceramic dental crowns INVENTOR(S): Hirabayashi, Masaya; Noda, Iwao

PATENT ASSIGNEE(S): Kyocera Corp., Japan SOURCE: Ger. Offen., 24 pp.

CODEN: GWXXBX

DOCUMENT TYPE: Patent LANGUAGE: German

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	DE 3435348	A1	19850502	DE 1984-3435348	19840926
	DE 3435348	C2	19860320		
	JP 60069007	.A	19850419	JP 1983-178962	19830926
PRIC	RITY APPLN. INFO.:			JP 1983-178962 A	19830926
AB	Dental crowns are o	composed	l of a glass	ceramic composed of	•
	microcrystals of Ca	a phosph	ate mineral	s, especially apatite,	composed of
	SiO2, P2O5, CaO, Mg	gO, and			
	CaF2 with a linear	expansi	on coeffici	ent of 9 + 10-6 to 11	
	+ 10-6/degree and a	a crysta	llization t	emperature of 800-900°.	The crown may
	be monolithic or a	composi	te of the g	lass ceramic and a meta	l or
	ceramic reinforceme	ent with	approx. th	e same expansion coeffi	cient The
	crowns may be used	to form	a bridge.	A mixture of SiO2 27,	
	P205 13, CaO 54, Mg	jO 3, в2	03 2, and		
	CaF2 1% was melted	in a Pt	crucible a	t 1450° and then	
	cooled rapidly to f	orm a g	lass cerami	c, which was melted at	
	1300° and cast in a	Mg pho	sphate-coat	ed centrifugal mold. T	he mold

cooled rapidly to form a glass ceramic, which was melted at 1300° and cast in a Mg phosphate-coated centrifugal mold. The mold and casting were heated at 800° for 2 h to crystallize a large portion of Ca10(PO4)60, and the mold was removed. The product had a linear expansion coefficient of 10 + 106/degree (800°), a bending strength of 1400 kg/cm2, and a compressive strength of 9000 kg/cm2. It was semitransparent, and an ivory segment could be fused to the inner surface. The crown was mounted on a prepared natural tooth with a glass ionomer cement.

L7 ANSWER 23 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1983:166924 CAPLUS

DOCUMENT NUMBER: 98:166924

TITLE: Crystallized glass for artificial bone

PATENT ASSIGNEE(S): Kyoto University, Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 5 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	JP 57191252	A	19821125	JP 1981-76696	19810522
	JP 62010939	В	19870309		
PRIO	RITY APPLN. INFO.:			JP 1981-76696	19810522
AB	Crystallized glasse	s for a	rtificial bo	ne and dental prosthetic	cs contain
	MgO 1-7, CaO 42-53,	SiO 22	-41 and P205	10-27%	
	with <10% of impuri	ties ind	cluding Li20	, Na2O, K2O, SrO, B2O3,	
	Al203, TiO2, ZnO2 N	b205, Ta	a205 and CaF	2. Thus, a crystallized	d glass
	consists of MgO 4.5	, CaO 42	2.0, SiO2 33	.5,	

P205 16.5 and CaF2 2.0% with a flexural strength of 1400 kg/cm2.

L7 ANSWER 24 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1969:516142 CAPLUS

DOCUMENT NUMBER: 71:116142

TITLE: Materials for coating articles

INVENTOR(S): McMillan, Peter W.; Partridge, Graham; Ward, Frank

Russell

PATENT ASSIGNEE(S): English Electric Co. Ltd.

SOURCE: Ger. Offen., 37 pp.

CODEN: GWXXBX

DOCUMENT TYPE: Patent LANGUAGE: German

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 1812733		19690904	DE 1968-1812733	19681204
FR 1593971			FR	
an 1005650			CD	

FR 1593971 FR
GB 1205652 GB
PRIORITY APPLN. INFO.: GB

PRIORITY APPLN. INFO.: GB 19671204

AB Glass compns. for coating, sealing, or joining articles, having a linear thermal expansion coefficient ≤50 + 10-7/degree consist of a molten mixture of a noncrystd. solder glass and 26-80% glass-ceramic having a linear thermal expansion coefficient ≤30 + 10-7/degree.

The glass-ceramic has a lower coefficient of expansion. The 2 components are present in such proportions that the composition has an

components are present in such proportions that the composition has an coefficient

of expansion equal or similar to that of the article. The solder glass contains PbO 30-80, B2O3 5-30, SiO2 0-5, ZnO 0-40, and BaO 0-30, or SiO2 35-66, ZnO 0-6, B2O3 0-19, BaO 0-21, Al2O3 0-15, CaO 0-4, Li2O 0-24, Na2O 0-13, K2O 0-29, MgO 0-4, BaF2 0-16, and CaF2 0-8. The glass-ceramic consists of SiO2 45-82, Al2O3 10-36, Li2O 0-25, and MgO 0-32, or SiO2 0-43, Al2O3 0-29, ZnO 21-70, and B2O3 14-58%, and as nucleation agents P2O5, V2O5, TiO2, or MoO3. The glass compns. can be used as coatings on different materials having a similar thermal expansion coefficient e.g. glass, ceramics, glass-ceramics, etc., to form decorative glazes or protective layers.

L7 ANSWER 25 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1968:98299 CAPLUS

DOCUMENT NUMBER: 68:98299

TITLE: Use of titanium- and vanadium-containing slags in the

production of silicate enamels

AUTHOR(S): Oshurkov, E. M.; Smirnov, N. S.; Antonova, S. N.

SOURCE: Trudy Ural'skogo Nauchno-Issledovatel'skogo Instituta

Chernykh Metallov (1966), 5, 259-67

CODEN: TUCMAN; ISSN: 0372-2465

DOCUMENT TYPE: Journal LANGUAGE: Russian

AB The possibility of using Ti-magnetite slags from the blast-furnace treatment of ores (slag A) and of slags from the treatment with calcined soda outside the blast-furnace of V-containing cast iron (slag B) for the preparation of enamels was studied. Slag A contains approx. melilite 40, baikovite (MgO, TiO2, SiO2) 15, perovskite 10, rutile

baikovite (MgO, TiO2, SiO2) 15, perovskite 10, rutile 3, FeO 2-3, and a glassy phase 20-35%. Slag B, that does not have a constant composition, contains, aside from Na metasilicates, thermonatrite (Na2CO3.H2O), Na2S, Na2SO4, and approx. 40% of a glassy phase, characteristic amts. of V2O5 1-1.5, Na2O 30-5, and Fe oxides 1.5-6.5. A D.T.A. was used to determine the softening points of the slags, alone or mixed with CaF2 or Na2SiF6, and to investigate the reactions with

other components of the enamels. By D.T.A. and a petrographical anal.,

ground enamels, containing the slags, were prepared The glass phases, present in both slags, accelerate the glass formation during melting of the enamels. The Fe oxides promote the fixation to a steel base. V205 lowers the surface tension of the melt, aids rheological properties, and causes wetting of oxidized steel surfaces. For example, a ground enamel was prepared from a mixture of slag B 42.55, quartz sand 25.25, CaF2 4.2, kaolin 5.4, boric acid 17.85, NaNO3 3.8, CoO 0.45, and NiO 0.5%. enamel obtained had the following composition: SiO2 45, TiO2 4.2, B203 11.9, Al203, 4, Na20 18.7, CaO 0.9, MgO 0.6, CaF2 4, P2O5 0.11, FeO 5.62, Fe2O3 1.88, Cr2O3 0.25, MnO 0.5, V2O5 0.73, Co2O3 0.55, Ni2O3 0.6, S 0.46%. The coefficient of thermal expansion between 20 and 460° was 65.8 + 10-7, the softening point was 400°. On coating steel with this enamel and submitting it to 8 temperature changes from 350 to 18°, did not show any damage, whereas the conventional enamel 2015/3132 did not stand 5 of such thermal changes without damage. Slag B was also used in enamels for coating Al.

ANSWER 26 OF 58 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1958:107827 CAPLUS

DOCUMENT NUMBER: 52:107827

ORIGINAL REFERENCE NO.: 52:19069g-i,19070a-c

TITLE:

Thin ceramic coatings resistant to shock and high temperatures for protecting sheet metals

INVENTOR (S): Long, John V.

PATENT ASSIGNEE(S): Solar Aircraft Co.

DOCUMENT TYPE: Patent Unavailable LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

APPLICATION NO. DATE KIND DATÉ PATENT NO. \_\_\_\_\_ \_\_\_\_\_ ----19580715 US 1953-377381 19530831 US 2843507

Protective ceramic coatings for sheet metals are formed from a AΒ mixture of ingredients selected from various groups according to the properties desired. The mixture, when ground, applied to the metal as an aqueous suspension, and fired, forms a crystalline or cryptocryst. coating

about the same crystal lattice as the underlying metal so that a good structural fit is attained, and forms also an intermediate layer of interdiffused base metal and ceramic coating, to promote adherence. The preferred ceramic ingredients have isometric, tetragonal, or hexagonal crystalline structures, and certain stated x-ray space groups. The mixture should include ingredients chosen from at least 2 of the following groups: (I) beryl, mullite, K feldspar, spodumene, Zn Zr silicate, Ca Zr silicate, Ba zirconate, for promoting stability and durability of the coating; (II) Si, Al, Cr, Fe, Ni, Co, Cr2O3, Fe3O4, Fe203, NiO, Co304, MnO, Cu20, for increasing adherence and resistance to thermal shock; (III) ZrO2, Al2O3, TiO2, ZrSiO4, SnO2, BeO, SiC, ZnO, for promoting durability of any glass phase formed; (IV) SiO2, B203, P205, to obtain more vitreous coatings; and (V) LiF, NaF, KF, CAF2, MgO, CaO, MoO2, as fluxes to make the coatings more impervious. Mixts., such as beryl about 50, Si 25, and SiC 25% require about 1 hr. of milling with water and a suspension agent, such as citric acid, Ca(NO3)2, methylcellulose, or the Na salt of polymerized D-mannuronic acid, before application to a base metal. Firing may be done at about 2200°F. for about 10 min. Coatings 0.002 in. thick protect sheet iron against oxidation at  $1600\,^{\circ}F.$  for extended periods, such as 120 hrs., with cyclic air quenching and soaking. Modified compns. have been derived by experiment for special uses through testing by thermal shock at temps. up to 2000°F, and other tests such as are used for vitreous enamelled metal samples. They may be 0.0005-0.003 in. thick and may be fired at 1500-2600°F. for 2-20 min., resp., preferably in N. Double

coatings may be applied when a vitreous, or an especially refractory, surface is desired. AISI type 314 steel was protected for 72 hrs. of cyclic testing at 1850°F. maximum temperature by firing first a coating of beryl 8, NiO 12, TiO2 6, and BeO 2 parts by weight at 2300°F. for 10 min. and secondly an outer coating of beryl 8, ZrO2 2, BeO 2, SiO2 8, B2O3 2, LiF 1, and NaF 2 parts by weight at 2250°F. for 10 min., the total thickness of both coatings being 0.00075 in.

#### COPYRIGHT 2007 EPO/FIZ KA on STN ANSWER 27 OF 58 EPFULL

2004:78856 EPFULL ACCESSION NUMBER:

20050323 ENTRY DATE PATENT: ENTRY DATE PUBLICATION: 20050323 UPDATE DATE PUBLICAT.: 20061124 DATA UPDATE DATE: 20061122 DATA UPDATE WEEK: 200647

TITLE (ENGLISH): Optical glass having a small photoelastic constant Verre optique a faible constante photoelastique TITLE (FRENCH):

TITLE (GERMAN): Optisches Glas mit niedriger fotoelastischer Konstante

INVENTOR(S): Junko, Ishioka, c/o Kabushiki Kaisha Ohara1-6-9,

Sagamihira, Sagamihara-shi, Kanagawa-ken, JP

KABUSHIKI KAISHA OHARA, 1-15-30, Oyama, PATENT APPLICANT(S):

Sagamihara-shi, Kanagawa-ken, JP PATENT APPL. NUMBER: 3120440

Joensson, Hans-Peter, et al, Patentanwaelte, von AGENT:

Kreisling Selting Werner, Deichmannhaus am Dom, 50667

A 20030916

Koeln, DE

AGENT NUMBER: 61904 DOCUMENT TYPE: Patent LANGUAGE OF FILING: English LANGUAGE OF PUBL.: English LANGUAGE OF PROCEDURE: English

German; English; French LANGUAGE OF TITLE:

EPA1 Application published with search report PATENT INFO TYPE:

JP 2003-323274

PATENT INFORMATION:

APPLICATION INFO.:

PRIORITY INFO.:

NUMBER KIND DATE EP 1516862 A1 20050323

DESIGNATED STATES: AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HU IE IT LI

> LU MC NL PL PT RO SE SI SK TR A 20040915 EP 2004-21888

ABEN

An optical glass having a small photoelastic constant  $(\beta)$  suitable for parts for polarizing optical system and light polarization control elements and having a refractive index (nd) within a range from 1.60 to 1.68 and an Abbe number (v d) within a range from 40 to less than 65 comprises, as atoms constituting the optical glass,

[[P] [5 - 10 mol % ]] [[Al] [1 - 3 mol % ]] [[Ba] [8 - 13 mol %, ]] [[Gd] [1 - 5 mol % ]] [[Nb] [0.1 - 3 mol % ]] [[F] [15 - 35 mol % and ]][[0] [40 - 52 mol %.]]

(image, 0.1, abstract drawing)

COPYRIGHT 2007 EPO/FIZ KA on STN L7 ANSWER 28 OF 58 **EPFULL** 

ACCESSION NUMBER: 2003:133768 EPFULL

ENTRY DATE PUBLICATION: 20051228 UPDATE DATE PUBLICAT .: 20061206 DATA UPDATE DATE:

20061206

DATA UPDATE WEEK:

200649

TITLE (ENGLISH): TITLE (FRENCH): TITLE (GERMAN):

BIOCOMPATIBLE MATERIAL MATERIAU BIOCOMPATIBLE BIOKOMPATIBLES MATERIAL

INVENTOR(S):

Martinez Fernandez, Julian, Universidad de

Sevilla, Pabellon de Brasil s/n, 41012 Sevilla, ES; Ramirez de Arellano Lopez, Antonio, Universidad de Sevilla, Pabellon de Brasil s/n, 41012 Sevilla, ES; Varela Feria, Francisco Manuel, Universidad de Sevilla, Pabellon de Brasil s/n, 41012 Sevilla, ES; Gonzalez Fernandez, Pio Manuel, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Serra Rodriguez, Julia A., Uni. de Vigo, ETS de

Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Liste Carmueja, Sara, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Chiussi, Stefano, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Perez Borrajo, Jacinto, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Arias Otero, Jose Luis, Uni. de

Vigo, ETS de Ingenieros Industriales, 36200 Vigo

(Pontevedra), ES; Leon Fong, Betty, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Perez-Martinez Y Perez-Amor, Mariano, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES

PATENT APPLICANT(S):

Universidad de Sevilla,, OTRI - Universidad de Sevilla, c/ Valparaiso, 5 1 planta, 41013 Sevilla, ES;

Universidad de Vigo, Officina I+D - Universidad de Vigo, Edificio CACTI, Campus Universitario de Vigo,

36200 Vigo (Pontevedra), ES

PATENT APPL. NUMBER:

2431773; 1896640

AGENT:

Temino Ceniceros, Ignacio, Abril Abogados, Amador de

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AGENT NUMBER: DOCUMENT TYPE: 159233 Patent Spanish

LANGUAGE OF FILING: LANGUAGE OF PUBL.:

English English

LANGUAGE OF PROCEDURE:

German; English; French

LANGUAGE OF TITLE:

EPA1 Application published with search report

PATENT INFO TYPE: PATENT INFORMATION:

PATENT INFORMATION:

NUMBER DATE KIND NUMBER KIND DATE \_\_\_\_\_\_ EP 1609441 A1 20051228 \_\_\_\_\_\_

DESIGNATED STATES:

WO 2004056292 20040708 AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HU IE IT LI

LU MC NL PT RO SE SI SK TR

APPLICATION INFO.:

EP 2003-785961 A 20031216 WO 2003-ES638 A 20031216 ES 2002-3052 A 20021220

PRIORITY INFO.:

# ABEN

The invention relates to a novel biocompatible material which is intended to be used in the production of implants, prostheses or biomedical devices, comprising biomorphic SiC ceramics as a support material with a coating of bioactive glass that is deposited by means of pulsed laser ablation.

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ACCESSION NUMBER:

2002:112098 EPFULL

DATA UPDATE DATE:

20031210

DATA UPDATE WEEK:

200350

TITLE (ENGLISH):

Biosoluble ceramic fiber composition with

improved solubility in a physiological saline solution

for a high temperature insulation material

TITLE (FRENCH):

Compositiond une ceramique biosoluble ayant une solubilite amelioree dans une solution saline

physiologique pour un materiau d isolation thermique

a haute temperature

TITLE (GERMAN):

Bioloesliche Keramikfaserzusammensetzung mit verbesserter Loeslichkeit in physiologischer

Salzloesung fuer einen Hochtemperaturwaermedaemmstoff Hong, Won Kak, 52-604 Karak Hyundai 5 cha Apt., 161-3

INVENTOR(S):

Karak-dong, Songpa-ku, Seoul, KR; Lee, Jae Wook,

302-1102 Shinan Apt., Kumi-dong, Boondang-ku, Sungnam, Kyunggi-do, KR; Kim, Kyung Hwan, No. 103, 128-3 Kumi-dong, Boondang-ku, Sungnam, Kyunggi-do, KR; Kim,

Jin Yeol, 403-209 Samik Apt., Sooseo-dong, Kangnam-ku, Seoul, KR; Kim, Hyung Sung, 311-1005 Joogong Apt.,

Doonchon-dong, Kangdong-ku, Seoul, KR

PATENT APPLICANT(S):

Kumkang Korea Chemical Co., Ltd., 1301-4 Seocho-dong,

Seocho-ku, Seoul, KR

PATENT APPL. NUMBER:

4329290

AGENT:

Fairbairn, Angus Chisholm, Marks & Clerk, 57-60

Lincoln's Inn Fields, London WC2A 3LS, GB

AGENT NUMBER: 81002

DOCUMENT TYPE:

Patent

LANGUAGE OF FILING: LANGUAGE OF PUBL.: English English

LANGUAGE OF PUBL.: LANGUAGE OF PROCEDURE:

English
German; English; French

LANGUAGE OF TITLE: PATENT INFO TYPE:

EPA3 Separate publication of search report

PATENT INFORMATION:

NUMBER KIND DATE

EP 1323687

A3 20031210

DESIGNATED STATES:

AT BE BG CH CY CZ DE DK EE ES FI FR GB GR IE IT LI LU

MC NL PT SE SI SK TR

EXTENSION STATES:

AL LT LV MK RO

APPLICATION INFO.:

EP 2002-258984

A 20021227

PRIORITY INFO.:

KR 2001-88372

A 20011229

# ABEN

The present invention relates to a biosoluble ceramic fiber composition with a superior solubility in a physiological saline solution for a high heat insulating material, and more particularly to the biosoluble ceramic fiber composition wherein the physiological saline solubility is improved by optimizing the amounts of network former (SiO2), network modifiers (CaO and MgO) and intermediates (ZrO2 and Al2O3), especially the physiological saline solubility at a high-viscosity range with a high proportion of network former (SiO2) is improved by optimizing the amounts of intermediates (ZrO2 and Al2O3); thermal and mechanical properties such as a heat resistance, a compressive strength and a restoring force are so improved as to be superior even at a high temperature by optimizing the amounts of network modifiers (CaO and MgO); and the conventional fiber forming equipments can be still employed to an economical advantage.

L7 ANSWER 30 OF 58 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER:

2002:42168 EPFULL

DATA UPDATE DATE: DATA UPDATE WEEK: 20040908 200437

TITLE (ENGLISH):

Spark plug

TITLE (FRENCH): Bougie d'allumage

TITLE (GERMAN): Zuendkerze

INVENTOR(S): Nishikawa, Kenichi, c/o NGK Spark Plug Co., Ltd., 14-18

Takatsuji-cho, Mizuho-ku, Nagoya-shi, Aichi, JP;

Sugimoto, Makoto, c/o NGK Spark Plug Co., Ltd., 14-18

Takatsuji-cho, Mizuho-ku, Nagoya-shi, Aichi, JP

PATENT APPLICANT(S): NGK SPARK PLUG CO., LTD, 14-18, Takatsuji-cho,

Mizuho-ku Nagoya-shi Aichi, JP

PATENT APPL. NUMBER: 560166

AGENT: Benson, John Everett, J. A. Kemp & Co., 14 South

Square, Gray's Inn, London WC1R 5JJ, GB

AGENT NUMBER: 47713
DOCUMENT TYPE: Patent
LANGUAGE OF FILING: English
LANGUAGE OF PUBL: English
LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE: German; English; French

PATENT INFO TYPE: EPA3 Separate publication of search report

PATENT INFORMATION:

NUMBER KIND DATE
----EP 1271725 A3 20040908

DESIGNATED STATES: AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU MC NL PT

SE TR

EXTENSION STATES: AL LT LV MK RO SI

APPLICATION INFO.: EP 2002-254502 A 20020626 PRIORITY INFO.: JP 2001-192668 A 20010626

#### ABEN

A spark plug comprising: a center electrode; a met shell; and an alumina ceramic insulator disposed between the center electrode and the metal shell, wherein at leastpart of the surface of the insulator is covered with a glaze layer comprising oxides, wherein the glaze layer comprises: 1 mol% or less of a Pb component in terms of PbO; 30 to 60 mol% of a si component in terms of SiO2; 20 to 50 mol% of a B component in terms of B2O3; 0.5 to 25 mol% of a Zn component in terms of ZnO; 0.5 to 15 mol% in total of at least one of Ba and Sr components in terms of BaO and SrO, respectively; 2 to 12 mol% in total of at least two alkaline metal components of Na, K and Li, in terms of Na2O, K2O, and Li2O, respectively, wherein K and Li is essential; and 0.1 to 10 mol% of a F component in terms of F2.

(image, 0.1, abstract drawing)

. L7 ANSWER 31 OF 58 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER: 2001:83627 EPFULL

DATA UPDATE DATE: 20041103
DATA UPDATE WEEK: 200445
TITLE (ENGLISH): Spark plug

TITLE (FRENCH): Bougie d'allumage

TITLE (GERMAN): Zuendkerze

INVENTOR(S): Nishikawa, Kenichi, c/o NGK Spark Plug Co. Ltd., 14-18

Takatsuji-cho, Mizuho-ku, Nagoya, Aichi, JP; Kouge,

Yoshihide, c/o NGK Spark Plug Co. Ltd., 14-18 Takatsuji-cho, Mizuho-ku, Nagoya, Aichi, JP

PATENT APPLICANT(S): NGK Spark Plug Company Limited, 14-18, Takatsuji-cho

Mizuho-ku, Nagoya, Aichi, JP

PATENT APPL. NUMBER: 560174

AGENT: Nicholls, Michael John, J.A. KEMP & CO. 14, South

Square Gray's Inn, London WC1R 5JJ, GB

AGENT NUMBER: 61941
DOCUMENT TYPE: Patent
LANGUAGE OF FILING: English
LANGUAGE OF PUBL.: English
LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE: German; English; French
PATENT INFO TYPE: EPB1 Granted patent

PATENT INFORMATION:

,	NUMBER	KIND	DATE
DESIGNATED STATES:	EP 1193816 DE FR GB IT	В1	20031112
APPLICATION INFO.: PRIORITY INFO.:	EP 2001-308311 JP 2000-299380		20010928
	JP 2001-244462		20000929
CITED PATENT LIT.:	EP 959542	Α	

## ABEN

A spark plug retains a resistor and has formed on alumina-based insulator a glaze layer, in which the glaze layer contains Pb component in a content of 1 mol% or less in terms of PbO, contains Si component, B component, Zn component, Al component, Ba component and/or Sr component, and contains F component in a content of 1 mol% or less. In addition, the glaze layer contains one kind or more of alkaline metal components, with Li component being necessary, and further contains one kind or more of phosphate ion, sulfate ion, fluoride ion and chloride ion in a content of 0.5 to 10 mol%. The glaze layer has a Vickers hardness Hv of 100 or more, shows excellent strength, especially impact resistance in spite of a reduced content of Pb component.

(image, 0.1, abstract drawing)

L7 ANSWER 32 OF 58 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER: 2001:47002 EPFULL

UPDATE DATE PUBLICAT.: 20050127
DATA UPDATE DATE: 20050126
DATA UPDATE WEEK: 200504
TITLE (ENGLISH): Spark plug

TITLE (FRENCH): Bougie d'allumage

TITLE (GERMAN): Zuendkerze

INVENTOR(S): Nishikawa, Kenichi, c/o NGK Spark Plug Co., Ltd., 14-18

Takatsuji-cho, Mizuho-ku, Nagoya-shi, Aichi, JP

PATENT APPLICANT(S): NGK SPARK PLUG CO., LTD, 14-18, Takatsuji-cho,

Mizuho-ku Nagoya-shi Aichi, JP

PATENT APPL. NUMBER: 560166

AGENT: Nicholls, Michael John, J.A. KEMP & CO. 14, South

Square Gray's Inn, London WC1R 5LX, GB

AGENT NUMBER: 61943
DOCUMENT TYPE: Patent
LANGUAGE OF FILING: English
LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE: German; English; French PATENT INFO TYPE: EPB1 Granted patent

DATENT THEORMATION:

FAIENT INFORMATION.	NUMBER	KIND	DATE
	EP 1168546	в1	20040204
DESIGNATED STATES:	DE FR GB IT		
APPLICATION INFO.:	EP 2001-305672	Α	20010629
PRIORITY INFO.:	JP 2000-197770	Α	20000630
CITED PATENT LIT.:	EP 959542	Α	
	US 5677250	Α	

# ABEN

The glaze layer 2d of the spark plug 100 includes oxides of: 15 to 60 mol% of a Si component in terms of SiO2; 22 to 50 mol% of a B component in terms of B2O3; 10 to 30 mol% of a Zn component in terms of ZnO; 0.5 to 35 mol% of Ba and/or Sr components in terms of BaO or

SrO; 1 mol% or less of an F component; 0.1 to 5 mol% of an Al component in terms of Al2O3; and 5 to 10 mol% in total of at least one of alkaline metal components of Na, K and Li, in terms of Na2O, K2O, and Li2, respectively, wherein Li is essential, and the amount of the Li component is 1.1 to 6 mol% in terms of Li2O.

(image, 0.1, abstract drawing)

L7 ANSWER 33 OF 58 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER: 2001:27728 EPFULL

ENTRY DATE PUBLICATION: 20050525
UPDATE DATE PUBLICAT.: 20061220
DATA UPDATE DATE: 20061220
DATA UPDATE WEEK: 200651

TITLE (ENGLISH): A METHOD FOR ETCHING THE SURFACE OF A BIOACTIVE

GLASS

TITLE (FRENCH): PROCEDE PERMETTANT DE GRAVER LA SURFACE D'UN VERRE

BIOACTIF

TITLE (GERMAN): VERFAHREN ZUR AETZUNG DER OBERFLAECHE EINES BIOAKTIVEN

GLASSES

INVENTOR(S): Itaelae, Ari, Itaeinen pitkaekatu 11 a 27, 20520 Turku,

FI; Aro, Hannu, Valtaojantie 4, 20810 Turku, FI; Hupa, Mikko, Rakuunatie 47, 20720 Turku, FI; Nordstroem, Egon, Norrskogsvaegen 3, 21600 Pargas, FI; Ylaenen, Heimo, Kunnallissairaalantie 64, 20810 Turku, FI

PATENT APPLICANT(S): Vivoxid Oy, Tykistoekatu 4 A, 20520 Turku, FI

PATENT APPL. NUMBER: 4287350

AGENT: Heikkilae, Hannes Antero, Turun Patenttitoimisto Oy,

P.O. Box 99, 20521 Turku, FI

AGENT NUMBER: 82342
DOCUMENT TYPE: Patent
LANGUAGE OF FILING: Finnish
LANGUAGE OF PUBL.: English
LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE: German; English; French
PATENT INFO TYPE: EPB1 Granted patent

PATENT INFORMATION:

PATENT INFORMATION:

NUMBER NUMBER	KIND KIND	DATE DATE
EP 1261556	B1 2	0050525
WO 2001066479	2	0010913

DESIGNATED STATES: AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU MC NL PT

SE TR

APPLICATION INFO.: EP 2001-913903 A 20010223

WO 2001-FI188 A 20010223 PRIORITY INFO.: FI 2000-515 A 20000307

CITED PATENT LIT.: WO 9621628 A1
WO 9706896 A1
US 4871384 A
US 4946546 A
US 5232878 A

L7 ANSWER 34 OF 58 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER: 2000:65657 EPFULL

UPDATE DATE PUBLICAT:: 20060810
DATA UPDATE DATE: 20060809
DATA UPDATE WEEK: 200632

TITLE (ENGLISH): NON-TOXIC, MICROBICIDAL DETERGENT TITLE (FRENCH): DETERGENT MICROBICIDE NON TOXIQUE

TITLE (GERMAN): NICHT-TOXISCHES, MIKROBIOZIDES REINIGUNGSMITTEL

INVENTOR(S): LEE, Sean, Oberlinstrasse 17, D-76227 Karlsruhe, DE

PATENT APPLICANT(S): Schott AG, Hattenbergstrasse 10, 55122 Mainz, DE

PATENT APPL. NUMBER: 2550661

AGENT: Fritzsche, Thomas, et al, Fuchs, Mehler, Weiss &

Fritzsche Naupliastrasse 110, 81545 Muenchen, DE

AGENT NUMBER: 60171
DOCUMENT TYPE: Patent
LANGUAGE OF FILING: German
LANGUAGE OF PUBL.: German
LANGUAGE OF PROCEDURE: German

LANGUAGE OF TITLE: German; English; French
PATENT INFO TYPE: EPB1 Granted patent

PATENT INFORMATION: PATENT INFORMATION:

PRIORITY INFO.:

NUMBER KIND DATE

NUMBER KIND DATE
----EP 1194518 B1 20040922

WO 2001004252 20010118

DESIGNATED STATES: AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU MC NL PT

SE

APPLICATION INFO.: EP 2000-956074 A 20000707

WO 2000-DE2230 A 20000707 DE 1999-19932238 A 19990709

CITED PATENT LIT.: US 4155870 A

L7 ANSWER 35 OF 58 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER: 2000:58304 EPFULL

ENTRY DATE PUBLICATION: 20050824
UPDATE DATE PUBLICAT.: 20060803
DATA UPDATE DATE: 20060802
DATA UPDATE WEEK: 200631

TITLE (ENGLISH): SILVER-CONTAINING, SOL-GEL DERIVED BIOGLASS

COMPOSITIONS

TITLE (FRENCH): COMPOSITIONS DE VERRE BIOACTIF CONTENANT DE L'ARGENT ET

OBTENU PAR UNE TECHNIQUE SOL-GEL

TITLE (GERMAN): SILBERHALTIGE BIOGLAS-ZUSAMMENSETZUNGEN, DIE VON

SOL-GEL ZUSTAENDEN ABGELEITET WERDEN

INVENTOR(S): BELLANTONE, Maria, 17 Cowley Street, London SW1F 3LZ,

GB; COLEMAN, Nichola J., 17 Purbeck Road, Chatham, Kent

ME4 6ED, GB; HENCH, Larry, L., 38 Clarence Gate

Gardens, London NW1 6BA, GB

PATENT APPLICANT(S): Imperial College Innovations, 47 Prince's Gate, London

SW7 2QA, GB

PATENT APPL. NUMBER: 3103210

AGENT: Clyde-Watson, Zoee, D Young & Co 120 Holborn, London

EC1N 2DY, GB

AGENT NUMBER: 126711
DOCUMENT TYPE: Patent
LANGUAGE OF FILING: English
LANGUAGE OF PUBL.: English
LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE: German; English; French
PATENT INFO TYPE: EPB1 Granted patent

PATENT INFORMATION: PATENT INFORMATION:

NUMBER KIND DATE

NUMBER KIND DATE
----EP 1196150 B1 20050824

WO 2000076486 20001221

DESIGNATED STATES: AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU MC NL PT

SE APPLICATION INFO.: EP 2000-939832 A 20000614 WO 2000-US16207 A 20000614 US 1999-139014P P 19990614 PRIORITY INFO.: PATENT ABSTRACTS OF JAPAN vol. 017, no. 659 (C-1137), 7 CITED NON PATENT LIT.: December 1993 (1993-12-07) & JP 05 213621 A (TOKUYAMA SODA CO LTD), 24 August 1993 (1993-08-24); PATENT ABSTRACTS OF JAPAN vol. 1997, no. 08, 29 August 1997 (1997-08-29) & JP 09 110463 A (ION KOGAKU SHINKO ZAIDAN), 28 April 1997 (1997-04-28); JORUNAL OF BIOMEDICAL RESEARCH, vol. 51, no. 3, 5 September 2000 (2000-09-05), pages 484-490, CITED PATENT LIT .: WO 8501210 Α WO 9404657 Α WO 9907777 Α US 5071674 Α US 5126141 Α US 5298260 Α US 5681872 Α US 5834008 Α US 5874101 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN L7 ANSWER 36 OF 58 ACCESSION NUMBER: 1998:72160 EPFULL ENTRY DATE PUBLICATION: 20050608 UPDATE DATE PUBLICAT.: 20060726 DATA UPDATE DATE: 20060726 DATA UPDATE WEEK: 200630 METHODS AND COMPOSITIONS FOR WHITENING TEETH TITLE (ENGLISH): METHODES ET COMPOSITIONS DE BLANCHIMENT DE DENTS TITLE (FRENCH): VERFAHREN UND ZUSAMMENSETZUNGEN ZUM BLEICHEN VON TITLE (GERMAN): ZAeHNEN LITKOWSKI, Leonard, J., 621 Sussex Road, Baltimore, MD INVENTOR(S): 21286, US; HACK, Gary, D., 8908 Skyrock Court, Columbia, MD 21046, US; GREENSPAN, David, C., 3116 North West 62nd Terrace, Gainesville, FL 32606, US University of Maryland, Baltimore, 520 W. Lombard PATENT APPLICANT(S): Street, Baltimore, Maryland 21201-1627, US; USBiomaterials Corporation, One Progress Boulevard, Box No.23, Alachua, FL 32615, US 594733; 2366000 PATENT APPL. NUMBER: Nash, David Allan, et al, HASELTINE LAKE, Redcliff Quay AGENT: 120 Redcliff Street, Bristol BS1 6HU, GB AGENT NUMBER: 59251 Patent DOCUMENT TYPE: English LANGUAGE OF FILING: English LANGUAGE OF PUBL.: LANGUAGE OF PROCEDURE: English LANGUAGE OF TITLE: German; English; French EPB1 Granted patent PATENT INFO TYPE: PATENT INFORMATION: PATENT INFORMATION: KIND DATE NUMBER NUMBER DATE KIND EP 1011621 B1 20050608 WO 9913852 19990325 DESIGNATED STATES: AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU MC NL PT SE A 19980918 APPLICATION INFO.: EP 1998-946860

CITED NON PATENT LIT.: EBISAWA Y. ET AL: 'Bioactivity of CaO.SiO2-based

WO 1998-US18500

US 1997-59222P

PRIORITY INFO.:

A 19980918 P 19970918

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glasses: in vitro evaluation' JOURNAL OF MATERIALS
                       SCIENCE: MATERIALS IN MEDICINE vol. 1, 1990, pages 239
                           OHURA K. ET AL: 'Bone-bonding ability of P2O5-free
                       CaO.SiO2 glasses' JOURNAL OF BIOMEDICAL MATERIALS
                       RESEARCH vol. 25, 1991, pages 357 - 365;
                           JOURNAL OF APPLIED BIOMATERIALS vol. 2, 1991, pages
                       231 - 239, XP008032929;
                           CALCIFIED TISSUE INTERNATIONAL vol. 57, 1995, pages
                       155 - 160, XP008032945
CITED PATENT LIT .:
                       EP 89136
                       WO 9610985
                                            Α
                       WO 9727148
                                            Α
                       WO 9951196
                                            Α
                       WO 9727148
                                            A1
                       US 4348381
                                            Α
                       US 4632826
                                            Α
                       US 5432130
                                            Α
                       US 5735942
                       EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN
      ANSWER 37 OF 58
L7
                       1997:79369 EPFULL
ACCESSION NUMBER:
UPDATE DATE PUBLICAT.: 20060621
DATA UPDATE DATE:
                       20060614
DATA UPDATE WEEK:
                       200624
                       BIOACTIVE GLASS COMPOSITIONS FOR
TITLE (ENGLISH):
                       USE IN THE TREATMENT OF TOOTH STRUCTURES
TITLE (FRENCH):
                       COMPOSITION DE VERRE BIOACTIF POUR UTILISATION POUR LE
                       TRAITEMENT DE STRUCTURES DENTAIRES
                       BIOAKTIVE GLASZUSAMMENSETZUNGEN ZUR VERWENDUNG ZUR.
TITLE (GERMAN):
                       BEHANDLUNG VON ZAHNSTRUKTUREN
                       LITKOWSKI, Leonard, J., 621 Sussex Road, Baltimore, MD
INVENTOR(S):
                       21286, US; HACK, Gary, D., 8908 Skyrock Court,
                       Columbia, MD 21046, US; GREENSPAN, David, C., 3116 N.W.
                       62nd Terrace, Gainesville, FL 32606, US
                       University of Maryland, Baltimore, 520 W. Lombard
PATENT APPLICANT(S):
                       Street, Baltimore, Maryland 21201-1627, US;
                       USBiomaterials Corporation, One Progress Boulevard, Box
                       No.23, Alachua, FL 32615, US
PATENT APPL. NUMBER:
                       594733; 2366000
                       Nash, David Allan, HASELTINE LAKE, Redcliff Quay 120
AGENT:
                       Redcliff Street, Bristol BS1 6HU, GB
AGENT NUMBER:
                       59251
DOCUMENT TYPE:
                       Patent
LANGUAGE OF FILING:
                       English
                       English
LANGUAGE OF PUBL.:
LANGUAGE OF PROCEDURE: English
                       German; English; French
LANGUAGE OF TITLE:
PATENT INFO TYPE:
                       EPB1 Granted patent
PATENT INFORMATION:
PATENT INFORMATION:
                       NUMBER ·
                                                  DATE
                                         KIND
                                                  DATE
                       NUMBER
                                         KIND
                       _____
                       EP 877716
                                   B1 20041013
                       ______
                       WO 9727148
                                               19970731
                       AT BE CH DE DK ES FI FR GB GR IE IT LI LU MC NL PT SE
DESIGNATED STATES:
                       RO SI
EXTENSION STATES:
                                            A 19970129
                       EP 1997-906489
APPLICATION INFO.:
                                           A 19970129
P 19960129
A 19960207
A 19970129
                       WO 1997-US1785
PRIORITY INFO.:
                       US 1996-10795P
                       US 1996-597936
                       US 1997-789909
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Encyclopedia of Chemical Technology, 3rd Ed., Vol. 21,
CITED NON PATENT LIT.:
                       pp 106-108
CITED PATENT LIT.:
                        EP 386525
                                             Α
                        EP 417018
                                             Α
                        EP 716049
                                             Α
                        WO 9610985
                                             Α
                        US 4057621
                        US 4239113
                        US 4605415
                        US 4775592
                                             Α
                        US 4775646
                        US 4783429
                                             Α
                        US 4851046
                                             Α
                        US 5120340
                                             Α
                        US 5204106
                                             Α
                        US 5429996
                                             Α
L7
       ANSWER 38 OF 58
                        EPFULL
                                  COPYRIGHT 2007 EPO/FIZ KA on STN
ACCESSION NUMBER:
                        1996:42343 EPFULL
DATA UPDATE DATE:
                        20030827
                        200335
DATA UPDATE WEEK:
                        MULTI-COLOR LIGHT EMISSION APPARATUS AND METHOD FOR
TITLE (ENGLISH):
                        PRODUCTION THEREOF
TITLE (FRENCH):
                        DISPOSITIF EMETTEUR DE LUMIERE EN PLUSIEURS COULEURS ET
                        PROCEDE DE PRODUCTION DE CE DISPOSITIF
TITLE (GERMAN):
                        VIELFARBIGE LICHTEMISSIONSVORRICHTUNG UND VERFAHREN ZUR
                        HERSTELLUNG DERSELBEN
INVENTOR(S):
                        EIDA, Mitsuru, Idemitsu Kosan Co., Ltd. 1280,
                        Kamiizumi, Sodegaura-shi Chiba-ken 299-02, JP;
                        MATSUURA, Masahide, Idemitsu Kosan Co., Ltd. 1280,
                        Kamiizumi, Sodegaura-shi Chiba-ken 299-02, JP;
                        TOKAILIN, Hiroshi, Idemitsu Kosan Co., Ltd. 1280,
                        Kamiizumi, Sodegaura-shi Chiba-ken 299-02, JP
PATENT APPLICANT(S):
                        IDEMITSU KOSAN COMPANY LIMITED, 1-1, Marunouchi 3-chome
                        Chiyoda-ku, Tokyo 100-0005, JP
PATENT APPL. NUMBER:
                        420820
                        Gille Hrabal Struck Neidlein Prop Roos, Patentanwaelte,
AGENT:
                        Brucknerstrasse 20, 40593 Duesseldorf, DE
AGENT NUMBER:
                        100973
DOCUMENT TYPE:
                        Patent
LANGUAGE OF FILING:
                        Japanese
LANGUAGE OF PUBL.:
                        English
LANGUAGE OF PROCEDURE:
                        English
                        German; English; French
LANGUAGE OF TITLE:
PATENT INFO TYPE:
                        EPB1 Granted patent
PATENT INFORMATION:
PATENT INFORMATION:
                        NUMBER
                                           KIND
                                                   DATE
                        NUMBER
                                           KIND
                        _____
                                            B1 20020904
                        EP 809420
                        ______
                        WO 9625020
                                                19960815
DESIGNATED STATES:
                        BE CH DE FR GB IT LI NL SE
APPLICATION INFO.:
                        EP 1996-901538 A 19960205
                        WO 1996-JP233
                                            A 19960205
                        JP 1995-41267
                                            A 19950206
PRIORITY INFO.:
                        JP 1995-49089
                                            A 19950214
                        JP 1995-299111
                                            A 19951024
CITED PATENT LIT .:
                        EP 387715
                                             Α
                        EP 550063
                                             Α
                        JP 2065094
                                             Α
                        JP 2082491
                                             Α
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JP 3280395

#### ABEN

This invention provides a multi-color light emission apparatus wherein a transparent inorganic oxide substrate (4) is disposed between an organic EL device (1) and a fluorescent layer (3) in such a manner as to arrange the fluorescent layer (3) with a gap with the organic EL device (1), and the organic EL device (1) is sealed by sealing means (5) between the transparent inorganic oxide substrate (4) and a support substrate (2). The invention provides also a multi-color light emission apparatus wherein a transparent insulating inorganic oxide layer (12) having a thickness of 0.01 to 200 µm is interposed between the fluorescent layer (3) and the organic EL device (1). In this way, light emission life and angle-of-view characteristics can be improved.

# (image, 0.1, abstract drawing)

T.7	ANSWER	39	OF	5.8	EPFIII.I.	COPYRIGHT	2007	EPO/FIZ	KΑ	on	STN

ACCESSION NUMBER: 1996:17574 EPFULL

DATA UPDATE DATE: 20000830 200035 DATA UPDATE WEEK:

Resistor for cathode ray tube and method of preparing TITLE (ENGLISH):

Resistance pour tube a rayons cathodique et methode TITLE (FRENCH):

pour sa production

Widerstand fuer Kathodenstrahlrohr und Verfahren zu TITLE (GERMAN):

dessen Herstellung

Aoki, Masaki, 5-12-1, Aoshinke, Minoo-shi, Osaka 562, INVENTOR(S):

> JP; Ohtani, Mitsuhiro, 3-8-31, Kouryounakamachi, Sakai-shi, Osaka 591, JP; Yamashita, Katuyoshi, 5-7-206, Myoukenzaka, Katano-shi, Osaka 576, JP;

Hiratani, Tatuo, 6639-1, Takayama-cho, Ikoma-shi, Nara 630-01, JP; Murai, Ryuichi, 4-4-67-105, Kamishinden,

Toyonaka-shi, Osaka 565, JP; Kudoh, Masatoshi,

39-5-507, Sakuragaoka-cho, Hirataka-shi, Osaka 573, JP;

Ohmae, Hideharu, 4-20-3, Higashitoyonakamachi,

Toyonaka-shi, Osaka 560, JP; Konda, Masahiko, 15-15, Takehashi-cho, Ibaraki-shi, Osaka 567, JP; Inoue, Mamoru, 1922-469, Noji-cho, Kusatsu-shi, Shiga 525, JP

PATENT APPLICANT(S): MATSUSHITA ELECTRIC INDUSTRIAL CO., LTD, 1006,

Oaza-Kadoma, Kadoma-shi, Osaka 571-0000, JP

PATENT APPL. NUMBER: 216884

Eisenfuehr, Speiser & Partner, Martinistrasse 24, 28195 AGENT:

Bremen, DE

AGENT NUMBER: 100151 DOCUMENT TYPE: Patent

LANGUAGE OF FILING: English LANGUAGE OF PUBL.: English

LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE: German; English; French PATENT INFO TYPE: EPB1 Granted patent

ΡΑΨΕΝΨ ΤΝΕΟΡΜΑΨΤΟΝ.

PATENT INFORMATION:	•		
	NUMBER	KIND	DATE
	EP 776868	 В1	19990915
DESIGNATED STATES:	DE		
APPLICATION INFO.:	EP 1996-119038	Α	19961128
PRIORITY INFO.:	JP 1995-309282	A	19951128
	JP 1996-37701	A	19960226
CITED PATENT LIT.:	EP 197584	A	
	EP 513909	A	
	US 4101708	A	
	US 4574055	A	
	US 4961022	A	
	US 4961023	A	

#### ABEN

The resistance layer formed on an inner face of the cathode ray tube has a resistance value with small dependency on sintering temperature and a superior characteristic of load and temperature.

The resistance layer contains RuO2 having a particle size of 0.05 to  $0.4~\mu\text{m}$  and the glass powder having a particle size of 0.01 to  $2.0~\mu\text{m}$ and if necessary metal oxides having a particle size of 0.05 to 2.0  $\mu m$ . The glass composition is preferably a PbO-B2O3-SiO2 -Al203-ZnO system.

(image, 0.1, abstract drawing)

ACCESSION NUMBER: 1995:61055 EPFULL

20020612 DATA UPDATE DATE: 200224 DATA UPDATE WEEK:

GLASS MATERIAL, SUBSTITUTION MATERIAL OF LIVING TISSUE TITLE (ENGLISH):

AND TEETH-STRAIGHTENING MATERIAL

MATERIAU VITREUX, MATERIAU DE SUBSTITUTION DE TISSUS TITLE (FRENCH):

VIVANTS ET MATERIAU DE REFORCEMENT DENTAIRE

GLASMATERIAL, ERSATZMATERIAL FUER LEBENDES GEWEBE UND TITLE (GERMAN):

ZAHNVERSTAERKENDES MATERIAL

NONAMI, Toru, c/o TDK Corporation 13-1, Nihonbashi INVENTOR(S):

1-chome, Chuo-ku Tokyo 103, JP; TAKAHASHI, Chihiro, c/o TDK Corporation 13-1, Nihonbashi 1-chome, Chuo-ku Tokyo

103, JP; SANO, Tatsuji, c/o TDK Corporation 13-1,

Nihonbashi 1-chome, Chuo-ku Tokyo 103, JP

TDK Corporation, 13-1, Nihonbashi 1-chome, Chuo-ku, PATENT APPLICANT(S):

Tokyo-to 103, JP

224160 PATENT APPL. NUMBER:

Adams, William Gordon, et al, RAWORTH, MOSS & COOK 36 AGENT:

Sydenham Road, Croydon Surrey CRO 2EF, GB

AGENT NUMBER: 27554 DOCUMENT TYPE: Patent Japanese LANGUAGE OF FILING: LANGUAGE OF PUBL.: English

LANGUAGE OF PROCEDURE: English LANGUAGE OF TITLE: German; English; French EPB1 Granted patent

PATENT INFO TYPE: PATENT INFORMATION:

PATENT INFORMATION:

EP 742019	B1 200	010620
	KIND KIND	DATE DATE

WO 9616683 19960606

DE FR GB DESIGNATED STATES:

APPLICATION INFO .: EP 1995-937200 A 19951128 WO 1995-JP2419 Α 19951128 PRIORITY INFO .: Α

JP 1994-321358 19941130 19950804 JP 1995-219849 Α

PATENT ABSTRACTS OF JAPAN vol. 013, no. 221 (C-598), 23 CITED NON PATENT LIT.:

May 1989 & JP 01 032867 A (SHIGERU SAITO; OTHERS: 01), 2

February 1989

EP 399787 CITED PATENT LIT .: Α

DE 4207180 Α JP 56007708 Α JP 62231668 Α JP 63082670 Α A glass material of the invention contains SiO2, MgO , Al2O3, and TiO2 as main components in amounts of 40-65% by weight of SiO2, 9-30% by weight of MgO, 8-31% by weight of Al2O3, and 6-15% by weight of TiO2, and satisfies expression I:  $\{100 - (A + S + T)\}/S \ge 0.340$  wherein A, S and T are contents in % by weight of Al2O3, SiO2, and TiO2, respectively, and Expression II: (S + M)/4 > 100 - (S + M + A + T) wherein M is a content in % by weight of MgO, and is substantially free of fluorine. It is used in the preparation of a living tissue replacement or orthodontic part. A living tissue replacement and orthodontic part which experience a minimal loss of material properties in a deleterious environment as in the oral cavity and should have a complex shape as in the case of a dental crown repair, high strength and aesthetic appearance can be briefly prepared in a safe manner without using a special manufacturing apparatus.

## L7 ANSWER 41 OF 58 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER: 1994:45234 EPFULL

UPDATE DATE PUBLICAT: 20051207
DATA UPDATE DATE: 20051207
DATA UPDATE WEEK: 200549

TITLE (ENGLISH): RESORBABLE BIOACTIVE PHOSPHATE CONTAINING CEMENTS

TITLE (FRENCH): CIMENTS CONTENANT DU PHOSPHATE BIOACTIF, CAPABLES DE SE

RESORBER

TITLE (GERMAN): RESORBIERBARE, BIOAKTIVE PHOSPAT-ZEMENTE

INVENTOR(S): Liu, Sung-Tsuen, 29 Landing, Laguna Niguel, California

92677, US; Chung, Harvey H., 43 Via Costa Verde, Rancho

DATE

Palos Verdes, California 92074, US

PATENT APPLICANT(S): Ceramedical, Inc.., 29 Landing, Laguna Niguel, CA

92677, US

PATENT APPL. NUMBER: 2320650

AGENT: Curtis, Philip Anthony, et al, A.A. Thornton & Co.,

235 High Holborn, London WC1V 7LE, GB

KIND

Α

AGENT NUMBER: 55274

DOCUMENT TYPE: Patent

LANGUAGE OF FILING: English

LANGUAGE OF PUBL.: English

LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE: German; English; French
PATENT INFO TYPE: EPB1 Granted patent

NUMBER

PATENT INFORMATION:

PATENT INFORMATION:

	NUMBER	KIND	DATE
	EP 729365	В1	20020731
	WO 9513835		19950526
DESIGNATED STATES:	AT BE CH DE DK ES	FR GB	IE IT LI NL.SE
APPLICATION INFO.:	EP 1994-909408	Α	19931115
	WO 1993-US11071	Α	19931115
PRIORITY INFO.:	EP 1994-909408	A	19931115 *
	WO 1993-US11071	A	19931115 *
CITED PATENT LIT.:	EP 324425	Α	
	EP 520690	Α	
•	EP 538914	Α	
	WO 9000892	Α	
	US 5149368	A	
	US 5218035	A	

L7 ANSWER 42 OF 58 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

US 5262166

ACCESSION NUMBER: 1994:11590 EPFULL

DATA UPDATE DATE: 19950308
DATA UPDATE WEEK: 199510

TITLE (ENGLISH): Canasite-apatite glass-ceramics
TITLE (FRENCH): Vitroceramiques de canasite-apatite

TITLE (GERMAN): Canasit-Apatit Glaskeramiken

INVENTOR(S): Wolcott, Christine Coulter, Corning Inc., Patent Dep.,

SP FR 02-12, Corning, NY 14831, US

PATENT APPLICANT(S): Corning Incorporated, Houghton Park, Corning New York

14831, US

PATENT APPL. NUMBER: 210458

AGENT: Smith, Sydney, et al, Elkington and Fife Prospect House

8 Pembroke Road, Sevenoaks, Kent TN13 1XR, GB

AGENT NUMBER: 36071
DOCUMENT TYPE: Patent
LANGUAGE OF FILING: English
LANGUAGE OF PUBL: English
LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE: German; English; French

PATENT INFO TYPE: EPA1 Application published with search report

PATENT INFORMATION:

NUMBER KIND DATE
EP 641556 A1 19950308

DESIGNATED STATES: DE FR GB IT

APPLICATION INFO.: EP 1994-111963 A 19940801 PRIORITY INFO.: US 1993-114759 A 19930901

#### ABEN

A glass-ceramic biomaterial having high strength and toughness, a family of glasses from which the glass-ceramic biomaterial can be produced, and a method of production. The material has a primary crystal phase of F-canasite and a secondary crystal phase of F-apatite. The glass family is SiO2-CaO-Na2O-K2O-P2O5-F. The method may be a single stage heat treatment, or a two stage involving an initial nucleation and a subsequent crystallization.

(image, 0.1, abstract drawing)

L7 ANSWER 43 OF 58 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER: 1994:7393 EPFULL

DATA UPDATE DATE: 20000308
DATA UPDATE WEEK: 200010

TITLE (ENGLISH): Glass material, living tissue replacement, method for

preparing living tissue replacement, and living tissue

replacement molding apparatus

TITLE (FRENCH): Materiau en verre, remplacement de tissu vivant,

methode de preparation d'un remplacement de tissu

vivant et installation pour le moulage de ce

remplacement

TITLE (GERMAN): Glasmaterial, Ersatz fuer lebendes Gewebe, Verfahren

zur Herstellung eines Ersatzes fuer lebendes Gewebe und

Anlage zum Formpressen dieses Ersatzes

INVENTOR(S): Nonami, Toru c/o TDK Corporation, 13-1, Nihonbashi

1-chome Chuo-ku, Tokyo, JP; Sano, Tatsuji c/o TDK Corporation, 13-1, Nihonbashi 1-chome Chuo-ku, Tokyo,

JP; Tsutsumi, Sadami 2, Morigamae-cho, Uzumasa,

Ukyo-ku, Kyoto-shi, Kyoto-fu, JP; Urabe, Tetsuo, 10-22, Shimeien Ibaraki-shi, Osaka-fu, JP; Fukuma, Masahiro, D32-304, 2-1, Karigaoka 9-chome, Hirakata-shi Osaka-fu,

JΡ

PATENT APPLICANT(S): TDK Corporation, 13-1, Nihonbashi 1-chome, Chuo-ku,

Tokyo-to 103, JP

PATENT APPL. NUMBER: 224160

AGENT: Vogeser, Werner, Dipl.-Ing., et al, Patent- und

Rechtsanwaelte Hansmann, Vogeser, Dr. Boecker, Alber, Dr. Strych, Liedl Albert-Rosshaupter-Strasse 65, 81369

Muenchen, DE

AGENT NUMBER: 12254 DOCUMENT TYPE: Patent LANGUAGE OF FILING: English LANGUAGE OF PUBL.: English LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE: German; English; French PATENT INFO TYPE: EPB1 Granted patent

PATENT INFORMATION:

	NUMBER	KIND	DATE	
	EP 626165	В1	19990317	
DESIGNATED STATES:	DE FR GB			
APPLICATION INFO.:	EP 1994-107633	Α	19940517	
PRIORITY INFO.:	JP 1993-139099	Α	19930517	
	JP 1993-214944	Α	19930806	
	JP 1993-353680	Α	19931229	
	JP 1994-80966	A	19940328	
CITED PATENT LIT.:	EP 231773	A		
	EP 401793	A		
	DE 4207180	Α		

#### ABEN

A living tissue replacement of crystallized glass having bioaffinity and mechanical strength is briefly obtained simply by pressure molding or machining without using a special equipment. A glass material having a softening point below its crystallization temperature and exhibiting viscous flow at temperatures below its melting point is heated at a temperature above its Tg and pressed at the temperature to mold to a desired shape, thereby manufacturing a living tissue replacement such as a dental crown. Molding can be done under a pressure of up to 20 MPa.

#### COPYRIGHT 2007 EPO/FIZ KA on STN L7 ANSWER 44 OF 58 EPFULL

1993:42684 EPFULL ACCESSION NUMBER:

20030219 DATA UPDATE DATE: DATA UPDATE WEEK: 200308

TITLE (ENGLISH): CEMENTS FROM BETA-DICARBONYL POLYMERS CEMENTS PRODUITS A PARTIR DE POLYMERES DE TITLE (FRENCH):

\$g(b)-DICARBONYLE

BETA-DICARBONYL POLYMERZEMENT TITLE (GERMAN):

MITRA, Sumita, B., Post Office Box 33427, Saint Paul, INVENTOR(S):

MN 55133-3427, US

MINNESOTA MINING AND MANUFACTURING COMPANY, (3m), 3M PATENT APPLICANT(S):

Center, P.O. Box 33427, St. Paul, Minnesota 55133-3427,

US

PATENT APPL. NUMBER: 300410

AGENT: Hammond, Andrew David, et al, Albihns Patentbyra

Goeteborg AB P.O.Box 142, 401 22 Goeteborg, SE

AGENT NUMBER: 74671 DOCUMENT TYPE: Patent English LANGUAGE OF FILING: LANGUAGE OF PUBL.: English English LANGUAGE OF PROCEDURE:

LANGUAGE OF TITLE: German; English; French PATENT INFO TYPE: EPB1 Granted patent

PATENT INFORMATION:

PATENT INFORMATION: .

NUMBER	KIND	DATE
NUMBER	KIND	DATE
EP 627907	В1	19970305

WO 9316676 19930902 DESIGNATED STATES: CH DE DK ES FR GB IT LI NL SE EP 1993-903089 A 19930115 APPLICATION INFO.: A 19930115 WO 1993-US380 A 19920227 US 1992-843420 PRIORITY INFO.: EP 313387 Α CITED PATENT LIT .: DE 4141174 Α GB 2213157 Α US 4209434 Α US 5017649 Α

L7 ANSWER 45 OF 58 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER: 1993:14049 EPFULL

DATA UPDATE DATE: 19940413 DATA UPDATE WEEK: 199415

TITLE (ENGLISH): Bone grafting material

TITLE (FRENCH): Materiau de remplacement pour les os

TITLE (GERMAN): Knochenersatzmaterial

INVENTOR(S): Kaneko, Norio, 35, Yoshida-machi, Totsuka-ku,

Yokohama-shi, Kanagawa-ken, JP

PATENT APPLICANT(S): NIKON CORPORATION, 2-3, Marunouchi 3-chome,

Chiyoda-Ku,, Tokyo, JP

PATENT APPL. NUMBER: 1099910

AGENT: DELETED Blumbach Weser Bergen Kramer Zwirner Hoffmann

Patentanwaelte R. 102(1) 31.12.1994, Radeckestrasse 43,

D-81245 Muenchen, DE

AGENT NUMBER: 100370
DOCUMENT TYPE: Patent
LANGUAGE OF FILING: English
LANGUAGE OF PUBL.: English
LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE: German; English; French

PATENT INFO TYPE: EPA1 Application published with search report

PATENT INFORMATION:

NUMBER KIND DATE

EP 591696 A1 19940413

DESIGNATED STATES: DE FR GB

APPLICATION INFO: EP 1993-114503 A 19930909

APPLICATION INFO.: EP 1993-114503 A 19930909 PRIORITY INFO.: JP 1992-271280 A 19921009

## ABEN

A bone grafting material for use in medicine is glass wool which has the mean diameter of 100  $\mu m$  or less and whose composition is:

[[SiO2] [40 - 62 % (w/w) ]] [[Na20] [10 - 32 % (w/w) ]] [[CaO] [10 - 32 % (w/w) ]] [[P205] [0 - 12 % (w/w) ]] [[CaF2] [0 - 12 % (w/w) ]] [[B2O3] [0 - 20 % (w/w) ]]

When said grafting material is used for treatment of periodontal disease, the grafting material is completely replaced by newly formed bone, wherein dental ankylosis of the grafting material to a tooth root does not occur and the newly formed bone and the tooth root are bound with a tissue like a periodontal membrane.

L7 ANSWER 46 OF 58 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER: 1990:29358 EPFULL

DATA UPDATE DATE: 19941221 DATA UPDATE WEEK: 199451

TITLE (ENGLISH): Magnetic memory storage devices

Dispositif magnetique de stockage d'information TITLE (FRENCH):

Magnetische Datenspeichervorrichtung TITLE (GERMAN):

Alpha, James William, 179 Watauga Avenue, Corning, New INVENTOR(S):

York 14830, US; Pinckney, Linda Ruth, 1546 Caton Road, Corning, New York 14830, US; Morgan, Walter Lane, 104

Weston Lane, Painted Post, New York 14870, US

Corning Incorporated, Houghton Park, Corning New York PATENT APPLICANT(S):

14831, US

PATENT APPL. NUMBER:

Smith, Sydney, et al, Elkington and Fife Prospect House AGENT:

8 Pembroke Road, Sevenoaks, Kent TN13 1XR, GB

AGENT NUMBER: DOCUMENT TYPE: Patent LANGUAGE OF FILING: English English LANGUAGE OF PUBL.: English LANGUAGE OF PROCEDURE:

LANGUAGE OF TITLE: German; English; French PATENT INFO TYPE: EPB1 Granted patent

PATENT INFORMATION:

NUMBER KIND DATE \_\_\_\_\_\_ EP 384574 B1 19941221

BE DE FR GB IT NL DESIGNATED STATES:

A 19900122 APPLICATION INFO.: EP 1990-300662 PRIORITY INFO.: US 1989-314883 A 19890224

GLASS AND CERAMICS. vol. 44, no. 7-8, August 1987, NEW CITED NON PATENT LIT.:

YORK US pages 286 - 290; Z.G. BEZSMERTNAYA ET AL: "A NEW

MECHANICALLY TREATED GLASS CRYSTAL MATERIAL"; PATENT ABSTRACTS OF JAPAN vol. 11, no. 268 (C-444) (2715) 29 August 1987, & JP-A-62 72547 (T.

MORISANE) 3 April 1987,

EP 47330 CITED PATENT LIT .: EP 222478

JP 6272547 Α JP 63210039 Α US 3689293 Α US 3905824 Α

US 4386162 Α US 4467039

ABEN

The present disclosure is related to the fabrication of a magnetic memory storage device consisting essentially of a head pad and a rigid information disk, which disk consists essentially of a rigid substrate having a coating of magnetic media on the surface thereof facing the head pad. More particularly, this disclosure is directed to a substrate for use in the disk. The substrate is selected from two different groups of glass-ceramics: the first group contains a sheet silicate as the predominant crystal phase; and the second group contains a chain silicate as the predominant crystal phase.

COPYRIGHT 2007 EPO/FIZ KA on STN L7 ANSWER 47 OF 58 EPFULL

ACCESSION NUMBER: 1990:10395 EPFULL

DATA UPDATE DATE: 19951004 DATA UPDATE WEEK: 199540

Use of ceramic materials for living hard TITLE (ENGLISH):

tissue replacements

Utilisation de ceramiques comme matieres de TITLE (FRENCH):

remplacement de tissu vivant

Verwendung von keramischen Materialen als TITLE (GERMAN):

Ersatzmaterialien fuer lebendes Gewebe

Nonami, Tohru, c/o TDK Corporation, 13-1 Nihonbashi INVENTOR(S):

1-chome, Chuo-ku, Tokyo, JP; Noma, Hiroyasu, 6-10-414 Tsudanuma 2-chome, Narashino-shi, Chiba, JP; Nakajima, Shinya, 12-10-606 Tsudanuma, 2-chome, Narashino-shi,

Chiba, JP

PATENT APPLICANT(S): TDK Corporation, 13-1, Nihonbashi 1-chome, Chuo-ku,

Tokyo-to 103, JP

PATENT APPL. NUMBER:

224160

AGENT:

Wey, Hans-Heinrich, Dipl.-Ing., et al, c/o Kanzlei

Hansmann, Vogeser und Partner Albert

Rosshaupter-Strasse 65, 81369 Muenchen, DE

AGENT NUMBER: 13062
DOCUMENT TYPE: Patent

LANGUAGE OF FILING: English
LANGUAGE OF PUBL.: English
LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE: German; English; French
PATENT INFO TYPE: EPB1 Granted patent

PATENT INFORMATION:

NUMBER KIND DATE \_\_\_\_\_\_ EP 401793 B1 19951004 DESIGNATED STATES: DE FR GB NL A 19900606 EP 1990-110716 APPLICATION INFO.: A 19890606 JP 1989-142058 PRIORITY INFO.: A 19900308 JP 1990-55126 JP 1990-131191 A 19900523 BIOMATERIALS, vol. 12, 1991; T. KOKUBO, pp. 155-163 CITED NON PATENT LIT.:

CITED NON PATENT LIT.: BIOMATERIALS, VOI. 12, 1991, 1. RO

GB 2080281 A
US 3926647 A

# ABEN

The materials for living hard tissue replacements of the present invention comprise a ceramic material which contain Cao and SiO2 as essential components, and MgO as an optional components, are substantially free from calcium phosphate and are capable of forming a compound based on calcium phosphate by contact with an aqueous solution containing phosphorus, and show high biological affinity.

# L7 ANSWER 48 OF 58 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER: 1989:36229 EPFULL

DATA UPDATE DATE: 19950830 DATA UPDATE WEEK: 199535

TITLE (ENGLISH): Heat-generating ceramics body for hyperthermia and

method of producing the same

TITLE (FRENCH): Corps ceramique thermogene pour l'hyperthermie et sa

methode de production

TITLE (GERMAN): Waermeerzeugender keramischer Koerper fuer Hyperthermie

und Verfahren zu seiner Erzeugung

INVENTOR(S): Kokubo, Tadashi, 46-1, Yokoyama Shimokaiinji,

Nagaokakyo City Kyoto-Fu, JP; Yamamuro, Takao, 100-33, Kitanokuchi Mozume-Cho, Muko City Kyoto-Fu, JP; Ohura, Koichiro 603, Takehana Heights, 6, Takehanajizoji-Minami-Machi, Yamashina-Ku Kyoto-City Kyoto-Fu, JP; Ebisawa, Yukihiro, 19-10, Tomio-Izumigaoka, Nara City

Nara Pref., JP

PATENT APPLICANT(S): KYOTO UNIVERSITY, 36 Yoshidahon-Machi Sakyo-ku, Kyoto

City Kyoto-Fu, JP

PATENT APPL. NUMBER: 621830

AGENT: Cresswell, Thomas Anthony, et al, J.A. KEMP & CO. 14

South Square Gray's Inn, London WC1R 5LX, GB

AGENT NUMBER: 50352

DOCUMENT TYPE: Patent
LANGUAGE OF FILING: English
LANGUAGE OF PUBL.: English
LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE: German; English; French

PATENT INFO TYPE:

EPB1 Granted patent

PATENT INFORMATION:

KIND DATE NUMBER \_\_\_\_\_ EP 361797 в1 19950830

DESIGNATED STATES:

DE FR GB IT NL .

APPLICATION INFO.: PRIORITY INFO.:

EP 1989-309645 A 19890921 JP 1988-238786 A 19880926

CITED NON PATENT LIT.:

INTERNATIONAL CONGRESS ON GLASS, Kyoto, 8th December 1974, part 2, pages 9-30 -9-41, Ceramic Society of Japan, Kyoto, JP; L.L. HENCH: "Biomedical applications

and glass corrosion"

CITED PATENT LIT .:

EP 40512 FR 2243915 US 4043821

#### ABEN

An excellent heat-generating ceramics body for hyperthermia of cancers is provided, comprising ferromagnetic ferrite particles, and a bioactive inorganic material layer surrounding the ferrite particles. The inorganic material layer has a composition to form apatite in a living body when embedded therein which has a high affinity to surrounding tissues of a living body and a property of not being perceived as a foreign material by the surrounding tissues by forming an apatite structure thereon, and the ferrite particles exhibits a highly efficient magnetic induction heat generation in an alternative magnetic field, without liberating noxious ions therefrom. The heat-generating ceramics body is particularly useful for deep cancers like bone tumors, can be applied in various shapes of powder, shaped bulk form or fiber depending on administration means, and can be used continuously or for a long period without an adverse influence. Methods of producing the same are also provided.

#### COPYRIGHT 2007 EPO/FIZ KA on STN ANSWER 49 OF 58 EPFULL L7

ACCESSION NUMBER:

1988:7329 EPFULL

DATA UPDATE DATE:

19910904

DATA UPDATE WEEK:

199136

TITLE (ENGLISH): TITLE (FRENCH):

Glass ceramics Vitroceramique

TITLE (GERMAN):

Glaskeramik

INVENTOR(S):

Takahashi, Katsuaki, 391 Hama Okayama City, JP; Miura, Yoshinari, 1155, Chikuma Kamigoricho, Akoh district Hyogo Prefecture, JP; Osaka, Akiyoshi, 416-1, Ohdaracho Okayama city, JP; Asada, Masayuki, 858-12, Mizue

Kurashiki city, JP

PATENT APPLICANT(S):

KURARAY CO., LTD., 1621 Sakazu, Kurashiki-City Okayama

Prefecture 710, JP

PATENT APPL. NUMBER:

298710

AGENT:

VOSSIUS & PARTNER, Postfach 86 07 67, 81634 Muenchen,

AGENT NUMBER: DOCUMENT TYPE:

100311 Patent

LANGUAGE OF FILING:

English English

LANGUAGE OF PUBL.: LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE:

German; English; French EPB1 Granted patent

PATENT INFO TYPE: PATENT INFORMATION:

> NUMBER KIND \_\_\_\_\_\_

EP 291013 B1 19910904 DESIGNATED STATES:

DE FR GB IT APPLICATION INFO.:

EP 1988-107494 A 19880510 EP 1988-107494 A 19880510 JP 1987-115434 A 19870511

PRIORITY INFO.: CITED NON PATENT LIT.: PATENT ABSTRACTS OF JAPAN, vol. 10, no. 335 (C-384)[2391], 13th November 1986; & JP-A-61 141 641

(HOYA CORP) 28-06-1986;

JOURNAL OF THE AMERICAN CERAMIC SOCIETY, vol. 67, no. 6, June 1984, pages C-100-C-101, Columbus, Ohio, US; S. KIHARA et al.: "Calcium phosphate glass-ceramic

crown prepared by lost-wax technique"

CITED PATENT LIT .:

DE 2808647 Α Α DE 3500287

## ABEN

A glass ceramic comprising at least 90 weight percent of a glass composed of

CaO: 17-28 weight percent P2O5: 13-26 weight percent A1203 : 25-38 weight percent

B2O3 : 20-37 weight percent (all based on the total weight of the glass ceramic),

with the atomic ratio of calcium to phosphorus within the range of 1.30 to 1.75, and apatite crystals dispersed as dominant crystals in said glass.

This glass ceramic is particulary useful as a biological material.

ANSWER 50 OF 58 COPYRIGHT 2007 EPO/FIZ KA on STN L7 EPFULL

ACCESSION NUMBER:

1987:29236 EPFULL

DATA UPDATE DATE:

19880316

DATA UPDATE WEEK:

198811

TITLE (ENGLISH): TITLE (FRENCH):

Synthesis of inorganic crystalline fibers Synthese de fibres cristallines minerales Synthese anorganischer kristalliner Fasern

TITLE (GERMAN): INVENTOR(S):

Flannery, James Elwin, 110 East Third Street, Corning

New York 14830, US; Wexell, Dale Richard, 372 West

Second Street, Corning New York 14830, US

PATENT APPLICANT(S):

Corning Glass Works, Sullivan Park FR-212, Corning New

York 14831, US 210453

PATENT APPL. NUMBER:

AGENT:

Froud, Clive, et al, Elkington and Fife Prospect House

8 Pembroke Road, Sevenoaks, Kent TN13 1XR, GB

AGENT NUMBER:

51991 Patent DOCUMENT TYPE: English LANGUAGE OF FILING: English LANGUAGE OF PUBL.:

LANGUAGE OF PROCEDURE:

English German; English; French

LANGUAGE OF TITLE: PATENT INFO TYPE:

EPA2 Application published without search report

PATENT INFORMATION:

PATENT INFORMATION:

	ÆER ÆER	KIND	DATE DATE
EP	260063	A2	19880316
EP	260063	A3	19890125

DESIGNATED STATES:

DE FR GB IT APPLICATION INFO.: EP 1987-307770 PRIORITY INFO.:

A 19870903 US 1986-904663 19860908

# ABEN

A inorganic crystalline fiber characterised in that it contains a minor amount of MoO3 and/or WO3 and/or As2O3 and wherein the predominant crystal phase is selected from fluormica, a fluoramphibole, canasite, potassium and/or sodium fluorrichterite, fluorapatite, and a fluoride-containing, spodumene-type crystal is disclosed, as is the production thereof.

ACCESSION NUMBER: 1987:553 **EPFULL** 

DATA UPDATE DATE: 19910424 DATA UPDATE WEEK: 199117

Mold, method of producing mold and casting method TITLE (ENGLISH): Moule, son procede de fabrication et procede de coulee TITLE (FRENCH):

TITLE (GERMAN): Giessform, Verfahren zur ihrer Herstellung und

Giessverfahren

Uchida, Seiju, 10-1-501, Wada 6-chome, Tamano-shi INVENTOR(S): Okayama, JP; Hashimoto, Akio, 625, Matsunami-Cho,

Takamatsu-Shi Kagawa, JP; Okuyama, Gen, 2-24-1, Chikko, Tamano-Shi Okayama, JP; Degawa, Toru, 2465-31, Amagi, Fujito-Cho, Kurashiki-Shi Okayama, JP; Sato, Takashi, 3-23, Nishitaga 1-Chome, Sendai-Shi.sctn.Kiyagi, JP; Fujiwara, Kozo, 8-24, Tai 2-Chome, Tamano-Shi Okayama,

MITSUI ENGINEERING & SHIPBUILDING CO., LTD, 6-4 Tsukiji PATENT APPLICANT(S):

5-chome, Chuo-ku, Tokyo 104, JP

PATENT APPL. NUMBER: 331440

Gruenecker, Kinkeldey, Stockmair & Schwanhaeusser AGENT: Anwaltssozietaet, Maximilianstrasse 58, 80538 Muenchen,

AGENT NUMBER: 100721 DOCUMENT TYPE: Patent English LANGUAGE OF FILING: LANGUAGE OF PUBL.: English LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE: German; English; French PATENT INFO TYPE: EPB1 Granted patent

PATENT INFORMATION:

FAIENT INFORMATION.	NUMBER		KIND	DATE
7767611777 671776		233478	В1	19910424
DESIGNATED STATES:	DE	FR GB IT SE	_	
APPLICATION INFO.: PRIORITY INFO.:		1987-100559	Α	19870116
		1986-7658	Α	19860117
	JP	1986-7659	Α	19860117
	JР	1986-7660	Α	19860117
·	JΡ	1986-7661	Α	19860117
	JР	1986-7662	A	19860117
	JР	1986-7663	Α	19860117
	JР	1986-7664	Α	19860117
	JР	1986-10916	Α	19860121
CITED PATENT LIT.:	DE	2047041	Α	
	GB	1346576	Α	
	US	1902419	Α	
	US	2201366	Α	
	US	2876122	A	
		3460606	A	
		511885	H	
	US	211002	п	

### ABEN

Disclosed are a porous calcia mold containing not less than 40 wt% CaO, a mold containing 95 to 10 wt% CaO and 5 to 50 wt% graphite, a mold containing not less than 40 wt% CaO, not more than 5 wt% low-eutectic temperature oxide and not more than 40 wt% high-eutectic temperature oxide, and a mold provided with a layer containing not less than 40 wt% CaO on the surface which comes into contact with a molten metal.

A method of producing a porous mold by baking a molded body obtained by a slip casting method or an injection molding method, and a method of producing a mold by baking a molded body (1) while bringing a member for preventing the deformation of a molded body (2) into contact with the molded

body (1) are also disclosed.

In addition, a method of casting a highly active metal, a high-melting point metal, or an alloy containing such a metal by using one of the above-described molds, and a casting method comprising the steps of placing a bottomless porous calcia mold on a metal chill plate, pouring a molten metal by top-pouring and gradually solidifying the molten metal from the lower part while insulating the riser are also disclosed.

## L7 ANSWER 52 OF 58 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER: 1986:27545 EPFULL

DATA UPDATE DATE: 19900131
DATA UPDATE WEEK: 199005

fluorrichterite

TITLE (FRENCH): Vitroceramique contenant de la cristobalite et de la

fluororichterite de potassium

TITLE (GERMAN): Cristobalit und Kalium-Fluorrichterit enthaltende

Glaskeramik

INVENTOR(S): Beall, George Halsey, 106 Woodland Drive, Big Flats New

York 14814, US; Pinckney, Linda Ruth, 103 Split Rail, Painted Post New York, US; Megles, John Edward, Jr., 78

Corning Boulevard, Corning New York, US

PATENT APPLICANT(S): Corning Glass Works, Sullivan Park FR-212, Corning New

York 14831, US

PATENT APPL. NUMBER: 210453

AGENT: Smith, Sydney, et al, Elkington and Fife Prospect House

8 Pembroke Road, Sevenoaks, Kent TN13 1XR, GB

AGENT NUMBER: 36071
DOCUMENT TYPE: Patent
LANGUAGE OF FILING: English
LANGUAGE OF PUBL.: English

LANGUAGE OF PROCEDURE: English

LANGUAGE OF TITLE: German; English; French

PATENT INFO TYPE: EPB1 Granted patent

PATENT INFORMATION:

NUMBER KIND DATE

EP 222478 B1 19900131

DESIGNATED STATES: AT BE DE FR GB IT NL

APPLICATION INFO.: EP 1986-307475 A 19860930

PRIORITY INFO.: US 1985-794841 A 19851104

CITED PATENT LIT.: GB 1307303 A

CITED PATENT LIT.: US 1985-794841 A
US 3839056 A
US 4467039 A

## ABEN

This invention is directed to the production of glass-ceramic articles exhibiting great toughness, a nodulus of rupture of at least 20,000 psi, (137,900 kPa) and, in the form of bars having dimensions 5"x0.5"x0.1", (12.70x1.27x0.254 cm) sagging during the crystallization heat treatment of no more than 0.2" (0.508 cm) over a 4" (10.16 cm) span, and wherein potassium fluorrichterite constitutes the predominant crystal phase but wherein cristobalite is also present in an amount of at least 10% by volume. The inventive articles having an overall composition consisting essentially, in weight percent on the oxide basis, of:

# (image, 18.1, tabular data)

L7 ANSWER 53 OF 58 MEDLINE on STN ACCESSION NUMBER: 2004127553 MEDLINE DOCUMENT NUMBER: PubMed ID: 15020110

TITLE: On the microstructure of biocomposites sintered from Ti, HA

and bioactive glass.

AUTHOR: Ning C Q; Zhou Y

CORPORATE SOURCE: Center for Biomedical Engineering, Wenner Gren Research

Laboratory, University of Kentucky, Lexington, KY 40506,

USA.. cqning@yahoo.com

SOURCE: Biomaterials, (2004 Aug) Vol. 25, No. 17, pp. 3379-87.

Journal code: 8100316. ISSN: 0142-9612.

PUB. COUNTRY: England: United Kingdom DOCUMENT TYPE: (EVALUATION STUDIES)

Journal; Article; (JOURNAL ARTICLE)

LANGUAGE: English

FILE SEGMENT: Priority Journals

ENTRY MONTH: 200410

ENTRY DATE: Entered STN: 16 Mar 2004

Last Updated on STN: 28 Oct 2004 Entered Medline: 27 Oct 2004

AB Sintering reactions and fine structures of the biocomposites prepared from

powder mixtures of titanium ( alpha -Ti), hydroxyapatite (HA) and

bioactive glass (BG) (SiO2-CaO-P2O5-B2O3-MgO-TiO2-CaF2) were

investigated by X-ray diffraction and transmission electron microscopy. The results showed that complex reactions among the starting materials mainly depended on the initial Ti/HA ratios as well as the sintering temperatures. And the reaction could be expressed by the following illustrative equation: Ti+Cal0(PO4)6(OH)2-->CaTiO3+CaO+TixPy+(Ti2O)+(Ca4P2O9)+H2O.

L7 ANSWER 54 OF 58 MEDLINE on STN ACCESSION NUMBER: 92305149 MEDLINE DOCUMENT NUMBER: PubMed ID: 1610964

TITLE: Influence of substituting B2O3 for CaF2

on the bonding behaviour to bone of glass-ceramics

containing apatite and wollastonite.

AUTHOR: Kitsugi T; Yamamuro T; Nakamura T; Yoshii S; Kokubo T;

Takagi M; Shibuya T

CORPORATE SOURCE: Department of Orthopaedic Surgery, Faculty of Medicine,

University of Kyoto, Japan.

SOURCE: Biomaterials, (1992) Vol. 13, No. 6, pp. 393-9.

Journal code: 8100316. ISSN: 0142-9612.

PUB. COUNTRY: ENGLAND: United Kingdom

DOCUMENT TYPE: Journal; Article; (JOURNAL ARTICLE)

LANGUAGE: English

FILE SEGMENT: Priority Journals; Space Life Sciences

ENTRY MONTH: 199207

ENTRY DATE: Entered STN: 7 Aug 1992

Last Updated on STN: 6 Feb 1998 Entered Medline: 30 Jul 1992

AB Glass-ceramics containing crystalline oxy-fluoroapatite (Ca10(PO4)6(O,F2)) and wollastonite (CaSiO3) (designated AWGC) are reported to have a fairly high mechanical strength as well as the capability of forming a chemical bond with bone tissue. The chemical composition is MgO 4.6,

CaO 44.9, SiO2 34.2, P2O5 16.3, and

CaF2 0.5 in weight ratio. In this study the influence of substituting B203 for CaF2 on the bonding behaviour of

glass-ceramics containing apatite and wollastonite to bone tissue was investigated. Two kinds of glass-ceramics containing apatite and

wollastonite were prepared. CaF2 0.5 was replaced with

B203 at 0.5 and 2.0 in weight ratio (designated AWGC-0.5B and AWGC-2.0B). Rectangular ceramic plates (15 x 10 x 2 mm, abraded

AWGC-2.0B). Rectangular ceramic plates (15 x 10 x 2 mm, abraded with Number 2000 alumina powder) were implanted into a rabbit tibia. The failure load, when an implant detached from the bone, or the bone itself broke, was measured. The failure load of AWGC-0.5B was 8.00 + -1.82 kg at 10 weeks after implantation and 8.16 + -1.36 kg at 25 weeks after implantation. The failure load of AWGC-2B was 8.08 + -1.70 kg at 10 weeks after implantation and 9.92 + -2.46 kg at 25 weeks after

implantation. None of the loads for the two kinds of glass-ceramics

decreased as time passed. Giemsa surface staining and contact microradiography revealed direct bonding between glass-ceramics and bone. SEM-EPMA showed a calcium-phosphorus rich layer (reaction zone) at the interface of ceramics and bone tissue. The thickness of the reaction zone was 10 to -15 microns and did not increase as time passed. (ABSTRACT TRUNCATED AT 250 WORDS)

L7 ANSWER 55 OF 58 BIOSIS COPYRIGHT (c) 2007 The Thomson Corporation on

STN

ACCESSION NUMBER: 2004:269602 BIOSIS DOCUMENT NUMBER: PREV200400266460

TITLE: On the microstructure of biocomposites sintered from Ti, HA

and bioactive glass.

AUTHOR(S): Ning, C. Q. [Reprint Author]; Zhou, Y.

CORPORATE SOURCE: Ctr Biomed EngnWenner Gren Res Lab, Univ Kentucky,

Lexington, KY, 40506, USA

cqning@yahoo.com

SOURCE: Biomaterials, (August 2004) Vol. 25, No. 17, pp. 3379-3387.

print.

CODEN: BIMADU. ISSN: 0142-9612.

DOCUMENT TYPE: Article LANGUAGE: English

ENTRY DATE: Entered STN: 26 May 2004

Last Updated on STN: 26 May 2004

AB Sintering reactions and fine structures of the biocomposites prepared from

powder mixtures of titanium (alpha-Ti), hydroxyapatite (HA) and

bioactive glass (BG) (SiO2-CaO-P2O5-B2O3-MgO-TiO2-CaF2) were

investigated by X-ray diffraction and transmission electron microscopy. The results showed that complex reactions among the starting materials mainly depended on the initial Ti/HA ratios as well as the sintering temperatures. And the reaction could be expressed by the following illustrative equation: Ti + Cal0(PO4)6(OH)2 fwdarw CaTiO3 + CaO

+ TixPy + (Ti20) + (Ca4P209) + H20. Copyright 2003 Elsevier Ltd. All

rights reserved.

L7 ANSWER 56 OF 58 SCISEARCH COPYRIGHT (c) 2007 The Thomson Corporation on

STN

ACCESSION NUMBER: 2006:121815 SCISEARCH

THE GENUINE ARTICLE: 00710

TITLE: Formation of hydroxyapatite onto glasses of the

CaO-MgO-SiO2 system with B2O3, Na2O, CaF2 and P2O5

additives

AUTHOR: Agathopoulos S; Tulyaganov D U; Ventura J M G; Kannan S;

Karakassides M A; Ferreira J M F (Reprint)

CORPORATE SOURCE: Univ Aveiro, CICECO, Dept Ceram & Glass Engn; P-3810193

Aveiro, Portugal (Reprint); Sci Res Inst Space Engn,

Tashkent 700128, Uzbekistan; Univ Ioannina, Dept Mat Sci &

Engn, GR-45110 Ioannina, Greece

jmf@cv.ua.pt

COUNTRY OF AUTHOR: Portugal; Uzbekistan; Greece

SOURCE: BIOMATERIALS, (MAR 2006) Vol. 27, No. 9, pp. 1832-1840.

ISSN: 0142-9612.

PUBLISHER: ELSEVIER SCI LTD, THE BOULEVARD, LANGFORD LANE,

KIDLINGTON, OXFORD OX5 1GB, OXON, ENGLAND.

DOCUMENT TYPE: Article; Journal

LANGUAGE: English

REFERENCE COUNT: 56

ENTRY DATE: Entered STN: 9 Feb 2006

Last Updated on STN: 9 Feb 2006

\*ABSTRACT IS AVAILABLE IN THE ALL AND IALL FORMATS\*

AB New bioactive glasses with compositions based on the CaO-

MgO-SiO2 system and additives of B2O3,

P2O5, Na2O, and CaF2 were prepared. The in vitro

mineralization behaviour was tested by immersion of powders or bulk glasses in simulated body fluid (SBF). Monitoring of ionic concentrations in SBF and scanning electron microscopy (SEM) observations at the surface of the glasses were conducted over immersion time. Raman and infrared (IR) spectroscopy shed light on the structural evolution occurring at the surface of the glasses that leads to formation of hydroxyapatite. (C) 2005 Elsevier Ltd. All rights reserved.

ANSWER 57 OF 58 SCISEARCH COPYRIGHT (c) 2007 The Thomson Corporation on

STN

ACCESSION NUMBER: -2004:330236 SCISEARCH

THE GENUINE ARTICLE: 808GS

On the microstructure of biocomposites sintered from Ti, TITLE:

HA and bioactive glass

**AUTHOR:** Ning C Q (Reprint); Zhou Y

CORPORATE SOURCE: Univ Kentucky, Ctr Biomed Engn, Wenner Gren Res Lab,

Lexington, KY 40506 USA (Reprint); Harbin Inst Technol,

Sch Mat Sci & Engn, Harbin 150001, Peoples R China

COUNTRY OF AUTHOR:

USA; Peoples R China

SOURCE:

BIOMATERIALS, (AUG 2004) Vol. 25, No. 17, pp. 3379-3387.

ISSN: 0142-9612.

PUBLISHER: ELSEVIER SCI LTD, THE BOULEVARD, LANGFORD LANE,

KIDLINGTON, OXFORD OX5 1GB, OXON, ENGLAND.

DOCUMENT TYPE:

Article; Journal

LANGUAGE:

English

REFERENCE COUNT:

26

ENTRY DATE:

Entered STN: 23 Apr 2004

Last Updated on STN: 23 Apr 2004

\*ABSTRACT IS AVAILABLE IN THE ALL AND IALL FORMATS\*

Sintering reactions and fine structures of the biocomposites prepared AB from powder mixtures of titanium (alpha-Ti), hydroxyapatite (HA) and

bioactive glass (BG) (SiO2-CaO-P2O5-B2O3-MgO-TiO2-CaF2) were

investigated by X-ray diffraction and transmission electron microscopy. The results showed that complex reactions among the starting materials mainly depended on the initial Ti/HA ratios as well as the sintering temperatures. And the reaction could be expressed by the following illustrative equation: Ti + Ca-10(PO4)(6)(OH)(2) --> CaTiO3 + CaO + TixPy + (Ti2O) + (Ca4P2O9) + H2O. (C) 2003 Elsevier Ltd. All rights reserved.

ANSWER 58 OF 58 SCISEARCH COPYRIGHT (c) 2007 The Thomson Corporation on 1.7

STN

1992:325243 SCISEARCH ACCESSION NUMBER:

THE GENUINE ARTICLE: HU926

TITLE:

INFLUENCE OF SUBSTITUTING B203 FOR CAF2

ON THE BONDING BEHAVIOR TO BONE OF GLASS-CERAMICS

CONTAINING APATITE AND WOLLASTONITE

KITSUGI T (Reprint); YAMAMURO T; NAKAMURA T; YOSHII S; AUTHOR:

KOKUBO T; TAKAGI M; SHIBUYA T

KYOTO UNIV, FAC MED, DEPT ORTHOPAED SURG, KYOTO 606, JAPAN CORPORATE SOURCE:

(Reprint); KYOTO UNIV, INST CHEM RES, MAT SCI LAB, KYOTO

606, JAPAN

COUNTRY OF AUTHOR:

**JAPAN** 

SOURCE:

BIOMATERIALS, (1992) Vol. 13, No. 6, pp. 393-399.

ISSN: 0142-9612.

PUBLISHER:

ELSEVIER SCI LTD, THE BOULEVARD, LANGFORD LANE,

KIDLINGTON, OXFORD, OXON, ENGLAND OX5 1GB.

DOCUMENT TYPE:

Article; Journal

FILE SEGMENT:

LIFE

LANGUAGE:

English

REFERENCE COUNT:

26

ENTRY DATE:

Entered STN: 1994

Last Updated on STN: 1994

\*ABSTRACT IS AVAILABLE IN THE ALL AND IALL FORMATS\*

=>

Glass-ceramics containing crystalline oxy-fluorapatite (Ca10(PO4)6(O,F2)) and wollastonite (CaSiO3) (designated AWGC) are reported to have a fairly high mechanical strength as well as the capability of forming a chemical bond with bone tissue. The chemical composition is MgO 4.6, CaO 44.9, SiO2 34.2, P205 16.3, and CaF2 0.5 in weight ratio. In this study the influence of substituting B203 for CaF2 on the bonding behaviour of glass-ceramics containing apatite and wollastonite to bone tissue was investigated. Two kinds of glass-ceramics containing apatite and wollastonite were prepared. CaF2 0.5 was replaced with B2O3 at 0.5 and 2.0 in weight ratio (designated AWGC-0.5B and AWGC-2.0B). Rectangular ceramic plates (15 x 10 x 2 mm, abraded with Number 2000 alumina powder) were implanted into a rabbit tibia. The failure load, when an implant detached from the bone, or the bone itself broke, was measured. The failure load of AWGC-0.5B was 8.00 +/-1.82 kg at 10 weeks after implantation and 8.16 +/- 1.36 kg at 25 weeks after implantation. The failure load of AWGC-2B was 8.08 +/- 1.70 kg at 10 weeks after implantation and 9.92 +/- 2.46 kg at 25 weeks after implantation. None of the loads for the two kinds of glass-ceramics decreased as time passed. Giemsa surface staining and contact microradiography revealed direct bonding between glass-ceramics and bone. SEM-EPMA showed a calcium-phosphorus rich layer (reaction zone) at the interface of ceramics and bone tissue. The thickness of the reaction zone was 10 to -15-mu-m and did not increase as time passed. There was no difference of the reaction zone thickness between AWGC-0.5B and AWGC-2.0B. The substitution B2O3 for CaF2 did not influence the bonding ability of the glass-ceramics. Boron neither promotes the dissolution of the glass-ceramics nor influences the bone formation at the interface of ceramics and bone.

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              AND CURRENT DISCOVER FILE IS DATED 25 SEPTEMBER 2006.
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         31755 L1 AND ((CALCIUM OXIDE) OR CAO)
=> s 12 and (silica or SiO2)
        17778 L2 AND (SILICA OR SIO2)
L3
=> s 13 and ((boron oxide) or B2O3)
          3885 L3 AND ((BORON OXIDE) OR B2O3)
=> s L4 and ((magnesium oxide) or MgO)
          2589 L4 AND ((MAGNESIUM OXIDE) OR MGO)
=> s L5 and ((calcium floride) or CaF2))
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number of left parentheses.
=> s L5 and ((calcium floride) or CaF2)
           111 L5 AND ((CALCIUM FLORIDE) OR CAF2)
=> s L6 and ((phosphorus pentoxide) or P2O5)
            58 L6 AND ((PHOSPHORUS PENTOXIDE) OR P2O5)
=> s 17 and biodegrad?
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=> d 18 1-3 ibib abs

ANSWER 1 OF 3 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

2004:182250 CAPLUS

DOCUMENT NUMBER:

140:205207

TITLE:

Biodegradable and bioactive

glass-ceramics, and method for fabricating the

same

INVENTOR(S):

Yu, Hyun Seung; Hong, Kug Sun; Kim, Hwan; Lee, Dong Ho; Lee, Choon Ki; Chang, Bong Soon; Kim, Deug Joong;

Seo, Jun Hyuk; Lee, Jae Hyup; Park, Ki Soo

PATENT ASSIGNEE(S):

S. Korea

SOURCE:

U.S. Pat. Appl. Publ., 17 pp.

CODEN: USXXCO

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

	PATENT NO.	KIND	DATE	APPLICATION NO.		DATE	
					-		
	US 2004043053	A1	20040304	US 2003-648217		20030827	
	KR 2004021111	Α	20040310	KR 2002-52566		20020902	
PRIO	RITY APPLN. INFO.:			KR 2002-52566	Α	20020902	
AB Disclosed herein is a biodegradable and bioactive							
glass-ceramic fabricated by mixing a slowly							
biodegradable glass-ceramic and a highly							

biodegradable glass-ceramic in a predetd. mixing ratio

wherein the bioactivity is maintained to be constant, and the biodegrdn.

rate is controlled by the mixing ratio. The biodegradable and

bioactive glass-ceramic is fabricated from a composition consisting of calcium oxide (CaO),

silica (SiO2), boron oxide (

B2O3), magnesium oxide (MgO), calcium fluoride (CaF2) and phosphorus

pentoxide (P2O5). Preparation of a biodegradable

and bioactive glass-ceramic according to

above method is disclosed. Electron microscopic images shown hydroxycarbonated apatite layers were formed on the entire surfaces of the specimens taken out 1 day after soaking in simulated body fluid thus suggesting that the specimens were highly bioactive. As the content of B203 in the specimens increased, the weight of the specimens was greatly reduced. This demonstrated that the biodegrdn. of the specimens had actively proceeded.

L8 ANSWER 2 OF 3 **EPFULL** COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER: 2003:133768 EPFULL

ENTRY DATE PUBLICATION: 20051228 UPDATE DATE PUBLICAT .: 20061206 DATA UPDATE DATE: 20061206 DATA UPDATE WEEK: 200649

TITLE (ENGLISH): BIOCOMPATIBLE MATERIAL TITLE (FRENCH): MATERIAU BIOCOMPATIBLE BIOKOMPATIBLES MATERIAL TITLE (GERMAN):

INVENTOR(S): Martinez Fernandez, Julian, Universidad de

Sevilla, Pabellon de Brasil s/n, 41012 Sevilla, ES; Ramirez de Arellano Lopez, Antonio, Universidad de Sevilla, Pabellon de Brasil s/n, 41012 Sevilla, ES; Varela Feria, Francisco Manuel, Universidad de Sevilla, Pabellon de Brasil s/n, 41012 Sevilla, ES; Gonzalez Fernandez, Pio Manuel, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Serra Rodriguez, Julia A., Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Liste Carmueja, Sara, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Chiussi, Stefano, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Perez Borrajo, Jacinto, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Arias Otero, Jose Luis, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Leon Fong, Betty, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES; Perez-Martinez Y Perez-Amor, Mariano, Uni. de Vigo, ETS de Ingenieros Industriales, 36200 Vigo (Pontevedra), ES

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PATENT APPL. NUMBER:

2431773; 1896640

AGENT:

Temino Ceniceros, Ignacio, Abril Abogados, Amador de

los Rios, 1 1, 28010 Madrid, ES

AGENT NUMBER:

159233

DOCUMENT TYPE:

Patent

LANGUAGE OF FILING:

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LANGUAGE OF PUBL.:

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LANGUAGE OF PROCEDURE:

English

LANGUAGE OF TITLE:

German; English; French

PATENT INFO TYPE:

EPA1 Application published with search report

PATENT INFORMATION:

PATENT INFORMATION:

NUMBER KIND DATE
NUMBER KIND DATE
EP 1609441 A1 20051228

DESIGNATED STATES:

AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HU IE IT LI

20040708

LU MC NL PT RO SE SI SK TR

WO 2004056292

APPLICATION INFO.:

EP 2003-785961 A 20031216 WO 2003-ES638 A 20031216 ES 2002-3052 A 20021220

PRIORITY INFO.:

ABEN

The invention relates to a novel biocompatible material which is intended to be used in the production of implants, prostheses or biomedical devices, comprising biomorphic SiC ceramics as a support material with a coating of bioactive glass that is deposited by means of pulsed laser ablation.

L8 ANSWER 3 OF 3 EPFULL COPYRIGHT 2007 EPO/FIZ KA on STN

ACCESSION NUMBER: 2000:58304 EPFULL

ENTRY DATE PUBLICATION: 20050824
UPDATE DATE PUBLICAT.: 20060803
DATA UPDATE DATE: 20060802
DATA UPDATE WEEK: 200631

TITLE (ENGLISH):

SILVER-CONTAINING, SOL-GEL DERIVED BIOGLASS

COMPOSITIONS

TITLE (FRENCH):

COMPOSITIONS DE VERRE BIOACTIF CONTENANT DE L'ARGENT ET

OBTENU PAR UNE TECHNIQUE SOL-GEL

TITLE (GERMAN):

SILBERHALTIGE BIOGLAS-ZUSAMMENSETZUNGEN, DIE VON

SOL-GEL ZUSTAENDEN ABGELEITET WERDEN

INVENTOR(S):

BELLANTONE, Maria, 17 Cowley Street, London SW1F 3LZ,

GB; COLEMAN, Nichola J., 17 Purbeck Road, Chatham, Kent ME4 6ED, GB; HENCH, Larry, L., 38 Clarence Gate Gardens, London NW1 6BA, GB Imperial College Innovations, 47 Prince's Gate, London PATENT APPLICANT(S): SW7 2QA, GB PATENT APPL. NUMBER: 3103210 Clyde-Watson, Zoee, D Young & Co 120 Holborn, London AGENT: EC1N 2DY, GB AGENT NUMBER: 126711 DOCUMENT TYPE: Patent LANGUAGE OF FILING: English LANGUAGE OF PUBL.: English LANGUAGE OF PROCEDURE: English LANGUAGE OF TITLE: German; English; French PATENT INFO TYPE: EPB1 Granted patent PATENT INFORMATION: PATENT INFORMATION: NUMBER KIND DATE NUMBER KIND -----EP 1196150 B1 20050824 -----WO 2000076486 20001221 DESIGNATED STATES: AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU MC NL PT SE APPLICATION INFO.: EP 2000-939832 A 20000614 WO 2000-US16207 A 20000614 P 19990614 PRIORITY INFO.: US 1999-139014P PATENT ABSTRACTS OF JAPAN vol. 017, no. 659 (C-1137), 7 CITED NON PATENT LIT.: December 1993 (1993-12-07) & JP 05 213621 A (TOKUYAMA SODA CO LTD), 24 August 1993 (1993-08-24); PATENT ABSTRACTS OF JAPAN vol. 1997, no. 08, 29 August 1997 (1997-08-29) & JP 09 110463 A (ION KOGAKU SHINKO ZAIDAN), 28 April 1997 (1997-04-28); JORUNAL OF BIOMEDICAL RESEARCH, vol. 51, no. 3, 5 September 2000 (2000-09-05), pages 484-490, CITED PATENT LIT .: WO 8501210 Α WO 9404657 Α WO 9907777 Α US 5071674 Α

US 5126141 Α US 5298260 Α US 5681872 Α US 5834008 Α US 5874101

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